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Karim Aissou, Hana Bouzit, Felix Krusch, Jean-Pierre Mericq, Didier Cot, et al.. Asymmetric Solvent-Annealed Triblock Terpolymer Thick Films Topped by a Hexagonal Perforated Lamellar Nanostructure. *Macromolecular Rapid Communications*, 2022, 43, pp.2100585. 10.1002/marc.202100585 . hal-03419294

HAL Id: hal-03419294

<https://cnrs.hal.science/hal-03419294v1>

Submitted on 8 Nov 2021

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Asymmetric Solvent-Annealed Triblock Terpolymer Thick Films Topped by a Hexagonal Perforated Lamellar Nanostructure

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Abstract: Asymmetric and nanostructured polystyrene-*block*-poly(2-vinyl pyridine)-*block*-poly(ethylene oxide) (PS-*b*-P2VP-*b*-PEO or SVEO, S:V:EO \approx 56:34:10, 79.5 kg.mol⁻¹) thick films blended with 20 wt % of a short PS homopolymer (hPS, 10.5 kg.mol⁻¹) were achieved by combining the non-solvent induced phase separation (NIPS) process with a solvent vapor annealing (SVA) treatment. Here, the NIPS step allows for the formation of a highly-permeable sponge-like substructure topped by a dense thin layer exhibiting poorly-ordered nanopores while the subsequent SVA treatment enables to reconstruct the material top surface into a porous monolayer of well-ordered hexagonal perforated lamellae (HPL). This optimized film architecture generated by NIPS-SVA showed a mean water permeability of 860 L h⁻¹ m⁻² bar⁻¹, which is roughly twice time higher than the flux measured through NIPS made PS-*b*-P2VP-*b*-PEO/hPS materials having poorly-ordered nanopores. The post-SVA treatment also revealed as a powerful tool to tailor the thickness of the nanostructure formed within the blended material since monoliths entirely composed of a HPL phase were produced by increasing the time of exposure to a chloroform stream. The water flux of such PS-*b*-P2VP-*b*-PEO/hPS monoliths was found to be an order of magnitude lower than that of their asymmetric film homologues.

Keywords: Solvent vapor annealing, block copolymer, self-assembly, asymmetric thick film, phase inversion process.

With their energy efficiency, manufacturing scalability, and ease of handling, ultrafiltration (UF) polymeric membranes are of particular interest in many applications such as protein separation and purification in the food and dairy industries, virus separation, and water treatment.¹⁻⁴ However, conventional UF membranes innately bear a permeability-selectivity trade-off,⁵ and are subject to fouling⁶ which is considered as one of the biggest challenges in separation technology. To overcome these issues, the development of novel

strongly selective and high-flux performance UF membranes endowed with fouling-resistant uniform nanopores is urgently needed.

To make progress towards the manufacture of such advanced materials, an ideal membrane architecture design, comprising a macroporous substructure topped by a thin layer of densely packed fouling-resistant and monodisperse nanopores, needs to be implemented at large scale. For that purpose, amphiphilic block copolymer (BCP) chains are considered as excellent candidates due to their innate tendency to self-assemble over large scale into a high areal density of nearly uniform nanofeatures with tailorable size, shape and chemistry.^{7,8}

Over the last decade, many efforts have been deployed to fabricate asymmetric and nanostructured BCP membranes devised with smart nanopores that can be potentially used to prevent the common problem of fouling. For instance, different external-stimuli responsive BCP membranes capable to adjust both the size and hydrophilic/hydrophobic character of their nanopores under changing environmental conditions (*e.g.*, pH and temperature) have been manufactured by self-assembly and non-solvent induced phase separation (SNIPS),⁸⁻¹⁴ which should allow facilitating the removal of foulants.¹⁵ To extend this concept, the nanodesign of smart BCP membranes having double-stimuli responsive pores has also been demonstrated.^{16,17}

Besides, as one of the most efficient methods to manufacture antifouling membrane is to weaken the interactions between the foulants and materials, the use of a hydrophilic block having a strong water affinity and large excluded volume is envisioned as a promising alternative to build fouling-resistant nanopores.¹⁸ In this direction, poly(ethylene oxide) (PEO)-containing AB- and ABC-type BCPs have been used to achieve asymmetric and nanostructured membranes with possibly improved fouling resistance and biocompatibility since PEO is well-known for its superior antifouling and low toxicity properties.¹⁹⁻²¹

In this work, we efficiently combine the non-solvent induced phase separation (NIPS) process and a solvent vapor annealing (SVA) treatment to produce PEO-containing ABC

triblock terpolymer thick films comprising a sponge-like substructure and a complex nanostructured top surface layer which differs from the common cylindrical phase generated by SNIPS. Indeed, we show that the disordered phase generated by NIPS on the top surface of blended polystyrene-*b*-poly(2-vinyl pyridine)-*b*-PEO (PS-*b*-P2VP-*b*-PEO) thick films is transformed into a monolayer of well-ordered hexagonal perforated lamellae (HPL) when exposed to a chloroform vapor for 45 min. This optimized material architecture with PEO-based nanochannels surrounded by a pH-responsive P2VP corona showed a water permeability value as high as $860 \text{ L h}^{-1} \text{ m}^{-2} \text{ bar}^{-1}$. Solvent-annealed (5h, CHCl_3) monoliths entirely composed of the HPL structure were also demonstrated. As the nanochannel tortuosity increases with the thickness of the HPL phase, blended PS-*b*-P2VP-*b*-PEO monoliths showed a lower water permeability ($30 \text{ L h}^{-1} \text{ m}^{-2} \text{ bar}^{-1}$).

Results

Figure 1a shows a typical square shaped (side $a = 3 \text{ cm}$) PS-*b*-P2VP-*b*-PEO thick film blended with 20 wt. % of hPS ($10.5 \text{ kg}\cdot\text{mol}^{-1}$) that was manufactured by NIPS from a 18 wt. % polymer solution in a di-solvent mixture of 1,4-dioxane and tetrahydrofuran (DOX/THF: 80/20 by weight). Importantly, the terpolymer thick films have been blended with small hPS chains to prevent the formation of cracks when the material is immersed in the water bath. This blended terpolymer thick film with a uniform dry thickness of $\sim 36 \mu\text{m}$ was produced by casting the polymer solution on a silicon substrate using a doctor blade with $250 \mu\text{m}$ gap, then by allowing the solvent to partially evaporate prior to plunge the film into a deionized water bath. As shown here, the resulted polymeric film dried at ambient conditions was deposited on a porous hydrophilic polyvinylidene difluoride (PVDF) support before its use in water flux tests (see the results discussed hereafter). The scanning electron microscopy (SEM) image presented in **Figure 1b** shows the representative top surface and substructure morphologies of a blended PS-

b-P2VP-*b*-PEO thick film generated by NIPS for a solvent evaporation time of 60s. In such conditions, the asymmetric thick film consists of a macroporous substructure morphology topped by a “dense” and nanoporous skin layer. It is noteworthy that an integral asymmetric thick film is generated by NIPS. Indeed, the bottom film layer, in contact with the silicon support before to be plunged in the water bath, mainly consists of an open network structure and big holes with no-evidence of preferential interactions of the PEO and P2VP domains with the silicon top surface (see **Fig. S1**). Both the magnified topographic atomic force microscopy (AFM) and SEM top view images presented in **Figures 1c-d** indicate that the pores generated on the material top surface form a poorly-ordered phase with a mean center-to-center spacing of ~43 nm, extracted from their respective fast transform Fourier (FFT) images. Note that the mean diameter size of the P2VP/PEO domains of ~19.1 nm (standard deviation, SD = 5.7 nm) does not correspond to the effective pore diameter since the P2VP-*b*-PEO chains partially fill their domains. Intriguingly, a well-ordered hexagonal array of P2VP/PEO features with a period of ~32.5 nm and mean domain diameter size of ~16.5 nm (SD = 2.3 nm) is randomly observed in the corner regions of the blended PS-*b*-P2VP-*b*-PEO thick films but this long-range ordered phase was not properly stabilized in this study (see **Fig. S2**). Such results are in accordance with previous studies demonstrating that the preparation of asymmetric and isoporous PS-*b*-P2VP-*b*-PEO thick films by SNIPS requires extensive optimization, due to the complex interplay of influencing parameters, such as solvent composition, polymer molecular weights, casting solution concentration and evaporation time.²²

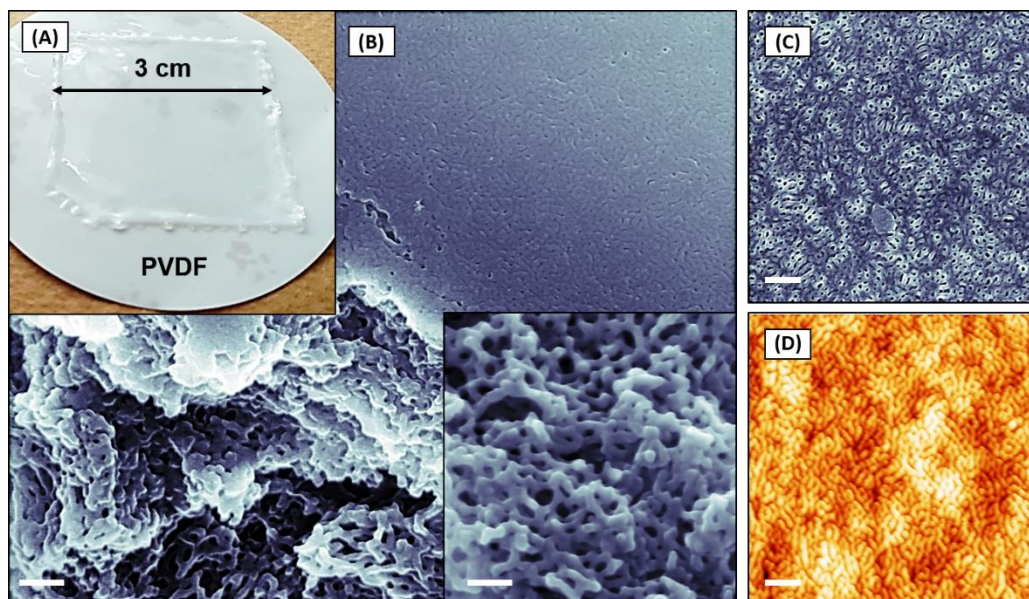


Figure 1: (a) Square shaped PS-*b*-P2VP-*b*-PEO/hPS thick film ($\sim 36 \mu\text{m}$) generated by NIPS then deposited on a porous hydrophilic PVDF support. (b) Top view SEM image of a NIPS-made PS-*b*-P2VP-*b*-PEO/hPS thick film revealing that the material architecture consists of a macroporous substructure topped by a dense and nanoporous skin layer. (c) and (d) Magnified top view SEM and AFM topographic images showing that the material top surface generated by NIPS exhibits poorly-ordered nanopores. Samples were treated by a CF_4 plasma prior both AFM and SEM imaging to enhance the contrast between PS and P2VP/PEO domains. Inset: Magnified SEM image showing that the macroporous substructure generated by NIPS adopts an open network structure. Scale bars: 250 nm.

To facilitate this optimization, Nunes and coworkers have recently proposed a semi-empirical method based on a guiding trend line calculated from the segregation strength analysis of the polymer-solvent mixtures supported by TEM imaging micelles in dilute solutions.²³ Here, we efficiently combine the NIPS process and the SVA approach (referred as NIPS-SVA)²⁴ to demonstrate that asymmetric thick films can also be manufactured with a very well nanostructured top surface derived from the self-assembly of PS-*b*-P2VP-*b*-PEO chains at equilibrium. **Figure 2** shows an asymmetric PS-*b*-P2VP-*b*-PEO/hPS thick film generated by NIPS-SVA then etched by a CF_4 plasma to enhance both SEM and AFM image contrasts. A well-ordered nanostructure is produced on the material top surface when the polymeric film is exposed to a CHCl_3 vapor during 45 min.

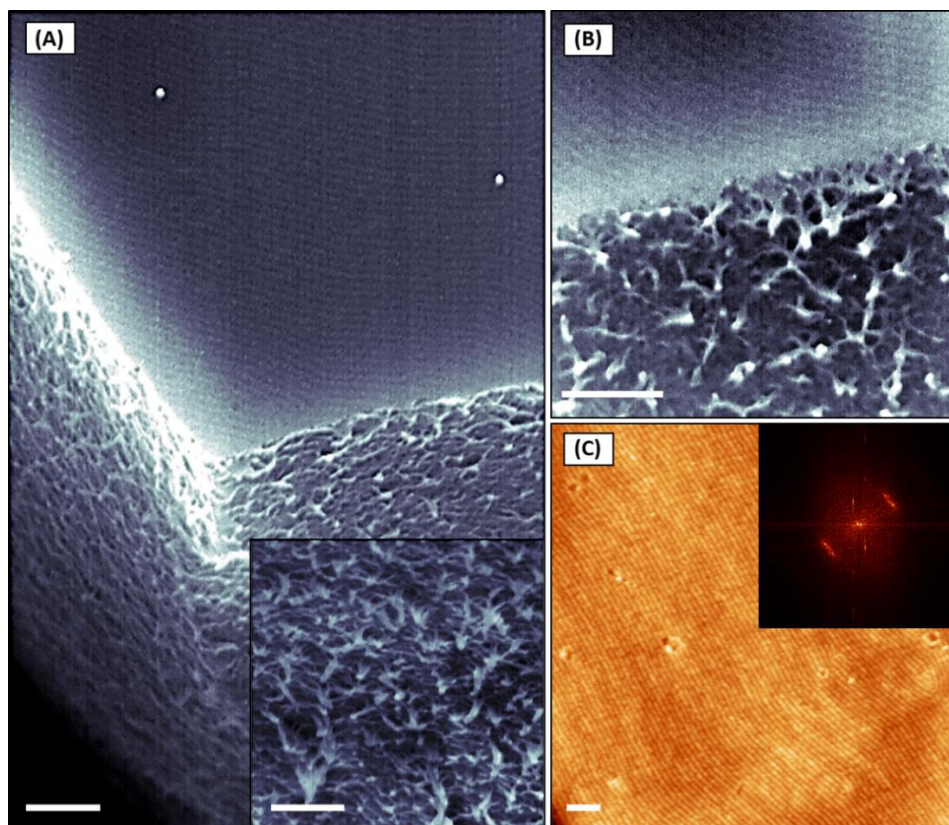


Figure 2: (a) and (b) Top view SEM images of a PS-*b*-P2VP-*b*-PEO/hPS thick film generated by NIPS then exposed to a CHCl₃ vapor for 45 min. Here, the solvent-annealed PS-*b*-P2VP-*b*-PEO/hPS thick film ($\sim 9 \mu\text{m}$) consists of a sponge-like substrate topped by a monolayer of OP lamellae. (c) AFM topographic image of a NIPS-SVA made (45 min, CHCl₃) PS-*b*-P2VP-*b*-PEO/hPS thick film revealing that a long-range-ordered lamellar nanostructure with a period of $\sim 38 \text{ nm}$, as extracted from the 2D-FFT, is produced on the material top surface. Samples were treated by a CF₄ plasma prior both AFM and SEM imaging to enhance the contrast between PS and P2VP/PEO domains. Inset: Magnified SEM image showing that the macroporous gutter layer generated by NIPS is transformed into a microporous (sponge-like) substructure after the SVA treatment. Scale bars: 250 nm.

The cross-sectional SEM views presented in **Figures 2a-b** show that the blended PS-*b*-P2VP-*b*-PEO thick film consists of a sponge-like substructure topped by an out-of-plane (OP) lamellar thin layer (sub-50 nm thick). Note that the formation of a denser sponge-like substructure during the SVA treatment makes that the thickness of the blended PS-*b*-P2VP-*b*-PEO film ($\sim 9 \mu\text{m}$) is now 4 times lower than that of the as-cast (no SVA) material. The 2D-FFT of the AFM topographic image displayed in **Figure 2c** indicates that the well-ordered top structure, consisting of an alternation of PS (bright) and P2VP/PEO (dark) lamellae, has a period of $\sim 38 \text{ nm}$. Although this pattern is the most frequently observed on the material-air interface, the

presence of another crystallographic plane oriented parallel to air surface indicates that the nanostructure is rather a hexagonal perforated lamellar phase. Indeed, the AFM topographic images presented in **Figure 3** show projections normal to the perforated layers where the PS lamellae and perforations appear bright while P2VP/PEO lamellae are dark after treating the terpolymer film with a CF₄ plasma.

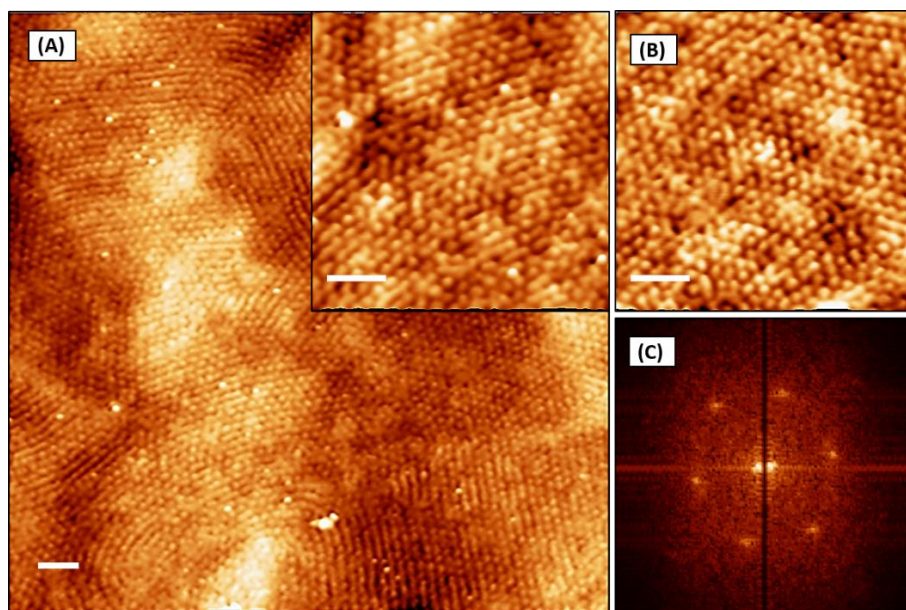


Figure 3: (a) AFM topographic image of a NIPS-SVA made (45 min, CHCl₃) PS-*b*-P2VP-*b*-PEO/hPS thick film treated by CF₄ plasma showing a hierarchical fingerprint pattern on the material top surface, composed of an alternation of PS (bright) and P2VP/PEO (dark) lamellae with PS bright perforations enclosed in the P2VP/PEO lamellae. (b) AFM topographic image and (c) its 2D-FFT revealing that the PS perforations formed within the P2VP/PEO lamellae are locally arranged into a hexagonal array with a period of ~38 nm. Inset: Magnified AFM topographic image of the PS-*b*-P2VP-*b*-PEO/hPS thick film top surface clearly showing the presence of bright dots arising from the PS perforations. Scale bars: 200 nm.

A “fingerprint pattern” formed by a lamellar structure, including some bright dots arising from PS perforations, can be observed from the **Figure 3a** while the AFM topographic view and its associated 2D-FFT displayed in **Figures 3b-c** indicate these perforations are locally arranged into a hexagonal array with a period of ~38 nm, in accordance with a HPL structure. Note that well-ordered PS perforations extended through P2VP/PEO lamellae are also observed on the material free surface along a crystallographic plane belonging to the {110} family, thereby confirming a high crystalline quality of the HPL structure formed on the sponge-like

substructure (see **Fig. S3**). To evidence that a nanophase-separation between the P2VP and PEO blocks should take place within the perforated-lamellar structure, differential scanning calorimetry (DSC) measurements were performed on the PS-*b*-P2VP-*b*-PEO powder (see **Fig. S4**). In addition to the superposition of the PS and P2VP glass transition temperatures occurring at $\sim 89^\circ\text{C}$, the heat flow curve reveals the melting point of PEO ($T_m \approx 49^\circ\text{C}$), which supports the P2VP and PEO blocks are nanophase-separated from each other as expected for triblock terpolymer chains having a linear topology. The formation of a tricolored pattern produced from self-assembled PS-*b*-P2VP-*b*-PEO chains was also reported by Huang *et al.*²⁵

To highlight the formation of a periodic and interconnected network morphology within the PS-*b*-P2VP-*b*-PEO/hPS thick films, the number of layers (*i.e.*, thickness) of the HPL structure has been increased by exposing the polymeric materials to a CHCl_3 vapor for longer time durations. The top view SEM image presented in **Figure 4a** shows that a 38 nm period fingerprint lamellar pattern is also produced at the material-air interface when the blended terpolymer thick film is placed under a CHCl_3 stream during 5h. In addition, the cross-sectional view SEM image of the full 9 μm thick PS-*b*-P2VP-*b*-PEO/hPS film reveals that the material is homogeneously dense across the whole film thickness (see **Fig. 4b**). Representative SEM images of the upper, middle and lower regions of a solvent-annealed (5h, CHCl_3) PS-*b*-P2VP-*b*-PEO/hPS layer confirm that the material asymmetry is not maintained when the SVA treatment is increased (see **Fig. 4c-e**). Indeed, a monolith entirely composed of the HPL structure is clearly evidenced by the presence of a periodic network phase extended from the air surface of the mesoporous material to the vicinity of the silicon substrate. It is noteworthy that the transformation of an asymmetric poly(1,1-dimethyl silacyclobutane)-*b*-PS-*b*-P2VP thick film, consisting of sponge-like substructure topped by a few hundred nanometer thick perforated lamellar (PL) structure, into an entirely PL-structured monolith has been recently

reported in the literature by increasing the SVA treatment duration, but no flux performance results were provided.²⁶

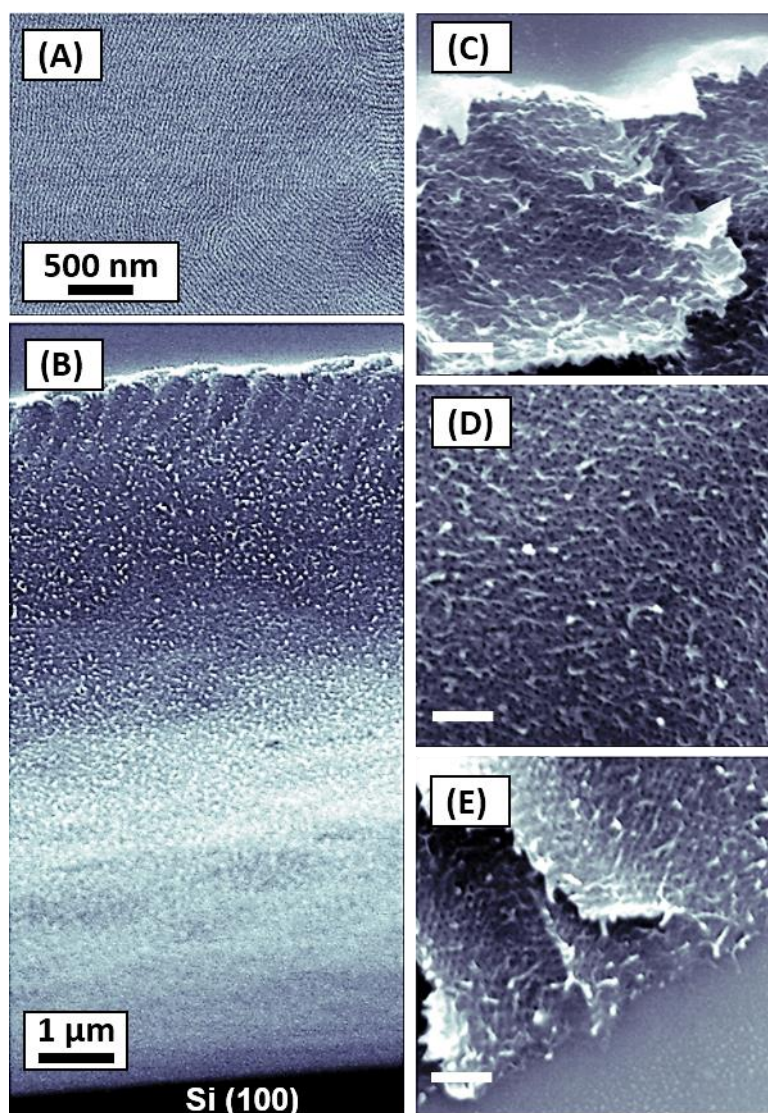


Figure 4: (a) Top view SEM image of a PS-*b*-P2VP-*b*-PEO/hPS thick film generated by NIPS then exposed to a CHCl₃ vapor for 5h from which a 38 nm period fingerprint lamellar pattern with a long-range order can be observed. Cross-sectional SEM views of a PS-*b*-P2VP-*b*-PEO/hPS monolith generated by NIPS-SVA (5h, CHCl₃) showing (b) the full material having a thickness of ~9 μm and the presence of a networked HPL nanostructure located (c) on the material air surface, (d) within the monolith and (e) in the vicinity of the Si substrate. Scale bars: 200 nm.

Here, to study the effect of the asymmetric material-to-monolith transformation on the material flux performances, the water permeability of several PS-*b*-P2VP-*b*-PEO/hPS thick films exposed to a CHCl₃ vapor during different times has been plotted as a function of water pressure applied from 0.5 to 1.5 bar, using a dead-end stirred ultrafiltration cell (see **Fig. 5**).

Regardless of the SVA treatment duration, all PS-*b*-P2VP-*b*-PEO/hPS thick films deposited on a porous (0.1 μm) hydrophilic PVDF support demonstrated an excellent stability over material failure at high feed pressure, as witnessed by a linear change in water flux with the increase in pressure drop.

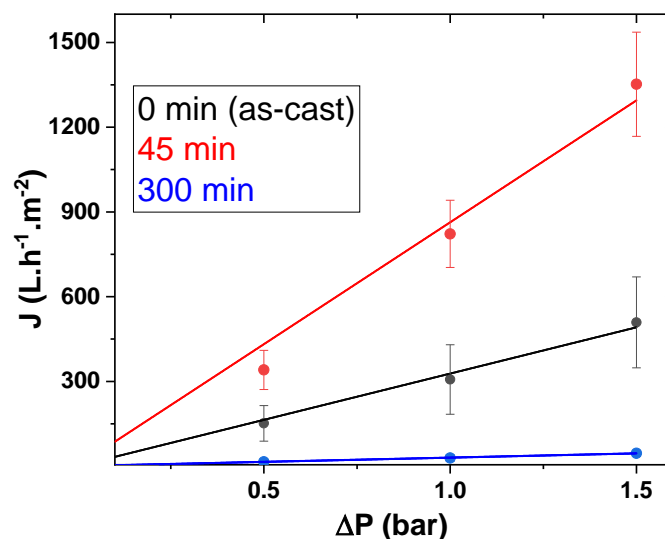


Figure 5: Water fluxes, J , of the NIPS made PS-*b*-P2VP-*b*-PEO/hPS materials exposed to a CHCl_3 vapor for different times: (black dots) 0 min, (red dots) 45 min and (blue dots) 300 min. Error bars are the standard deviations of the mean relative permeability values calculated for 3 different samples.

The as-cast (no SVA) PS-*b*-P2VP-*b*-PEO/hPS thick film having a macroporous substructure showed a permeability of $330 \text{ L h}^{-1} \text{ m}^{-2} \text{ bar}^{-1}$ which is roughly 2.5 times smaller than the permeability through a solvent-annealed (45 min, CHCl_3) HPL thick film having a (sponge-like) microporous substructure ($860 \text{ L h}^{-1} \text{ m}^{-2} \text{ bar}^{-1}$). To confirm that such an increase of the water transport is not dominated by the formation of macro defects (such as “cracks”) within the PS-*b*-P2VP-*b*-PEO/hPS thick film during the SVA treatment, the rejection of 10 nm gold nanoparticles (NPs) is shown in the **Figure S5**. A rejection of $\sim 73\%$ was achieved at 1 bar for the NIPS-SVA made (45 min, CHCl_3) PS-*b*-P2VP-*b*-PEO/hPS material suggesting that the water transport mainly occurs through the perforated-lamellar PEO-based nanodomains. Conversely, a colorless permeate with no absorption at 520 nm in the UV-visible spectrum can be achieved from as-cast PS-*b*-P2VP-*b*-PEO/hPS thick films, implying that the 10 nm gold NPs

are completely blocked and, therefore, that the poorly ordered P2VP/PEO pores have their dimension below 10 nm (see **Fig. S6**).

The increase of the water transport through the solvent-annealed polymeric thick film is probably due mainly to the variation of the area occupied by the porous P2VP/PEO domains formed on the material skin layer since an OP lamellar morphology improves the PS/(P2VP/PEO) matrix/pore area ratio at the free surface in comparison with poorly-ordered nanostructures, even if they are situated on top of a gutter layer with macropores. In other words, this behavior is in accordance with dimensionality reduction considerations implying that the water transport properties through cylinders are lower than that through lamellae.²⁷ It is also noteworthy that the columnar-to-lamellar (rearrangement) transition occurring during the SVA treatment induces an increase of the PEO-based domain size that should be taken into account in the water permeability variation. This increase of the PEO-based domain size, evidenced by the gold NP retention tests, is also confirmed by the performance of bovine serum albumin (BSA) filtrations at pH of 7 (0.5 g/L of BSA in Milli-Q ultrapure water) since protein rejections of ~69 % and ~44 % were achieved at 1 bar from the NIPS and NIPS-SVA made (45 min, CHCl₃) PS-*b*-P2VP-*b*-PEO/hPS thick films, respectively. For comparison, a rejection of ~67% of BSA was demonstrated (at 0.5 bar and pH = 7.4) by the Peinemann group from asymmetric SNIPS-made PS-*b*-PEO membranes, consisting of a nanostructured top surface with well-ordered OP PEO cylindrical domains and a macroporous substructure (water fluxes around 800-850 L h⁻¹ m⁻² bar⁻¹).²⁸

By increasing the thickness of the HPL structure within the PS-*b*-P2VP-*b*-PEO/hPS thick films from few tens of nanometers to several micrometers, the water flux is drastically decreased. Indeed, by exposing the polymeric materials to a CHCl₃ vapor during 2h, a thicker HPL-forming top layer is produced on the asymmetric PS-*b*-P2VP-*b*-PEO/hPS films, which conduces to a lower water permeability of 230 L h⁻¹ m⁻² bar⁻¹ (see **Fig. S7**). This phenomenon

is more marked for the solvent-annealed (5h, CHCl₃) HPL monoliths (see Fig. 4) since these symmetric films exhibit a water permeability of 30 L h⁻¹ m⁻² bar⁻¹, which is an order of magnitude lower compared to the maximized value achieved from asymmetric NIPS-SVA made (45min, CHCl₃) PS-*b*-P2VP-*b*-PEO/hPS materials having a HPL monolayer (see **Fig. 5**). According the Hagen-Poiseuille law describing the relationship between the flux value and the tortuosity and thickness of the nanostructured layer, this result seems qualitatively consistent with a superior flux performance for the asymmetric PS-*b*-P2VP-*b*-PEO/hPS material having a lower porous nanochannel tortuosity due to the formation of a thinner HPL nanostructure.

Conclusion

Fouling-resistant PEO-containing terpolymer thick films with an advantageous architecture design, comprising a microporous substructure and a nanostructured porous top surface layer, has been achieved by using a simple and scalable methodology combining the conventional NIPS process with a SVA treatment used in advanced lithography application. By using this methodology, blended PS-*b*-P2VP-*b*-PEO thick films, showing a well-defined nanoporous perforated lamellar monolayer situated on top of a gutter layer, have been devised to provide an excellent water permeability value (860 L h⁻¹ m⁻² bar⁻¹). Such kind of asymmetric and nanostructured materials endowed with PEO-based porous nanochannels surrounded by a pH-responsive P2VP corona are highly desired to move towards the next-generation of strongly selective and high flux performance UF membranes having potential antifouling capabilities. However, as the perforated lamellar morphology generated by NIPS-SVA showed a lower protein rejection than that of the poorly ordered columnar phase manufactured by NIPS, we are currently working on the formation of a core-shell double gyroid structure with PEO-based nanochannels topping a sponge-like substructure in order to improve the selectivity of PS-*b*-P2VP-*b*-PEO made materials by reducing the pore dimensionality.

Methods

Materials: Both the PS homopolymer ($10.5 \text{ kg}\cdot\text{mol}^{-1}$) with a dispersity, D , of 1.09 and PS-*b*-P2VP-*b*-PEO terpolymer, consisted of PS ($S = 45 \text{ kg}\cdot\text{mol}^{-1}$), P2VP ($V = 26 \text{ kg}\cdot\text{mol}^{-1}$) and PEO ($EO = 8.5 \text{ kg}\cdot\text{mol}^{-1}$) (volume fraction ratios, S:V:EO = 56:34:10) and having D of 1.05, were purchased from Polymer Source Inc., Canada. All the solvents: 1,4-dioxane (DOX, >99%), tetrahydrofuran (THF, >99%) and chloroform (CHCl_3 , >97%) were purchased from Sigma-Aldrich and used as received. ($0.1 \mu\text{m}$, 47 mm) Hydrophilic Durapore PVDF supports (Merck Milipore) were purchased from Sigma-Aldrich and used as received.

Fabrication of asymmetric PS-*b*-P2VP-*b*-PEO/hPS thick films by NIPS: $36 \mu\text{m}$ thick PS-*b*-P2VP-*b*-PEO thick films blended with 20 wt % of hPS were drawn onto (3 x 3 cm) silicon substrates by using a simple tape casting technique with $250 \mu\text{m}$ gap from a 18 % wt. terpolymer solution in a di-solvent mixture (DOX/THF: 80/20 by weight). To produce asymmetric terpolymer thick films, solvents were allowed to evaporate during 60s at room temperature (RT) to form a dense air surface layer with a kinetically trapped nanoporous structure, and then the blended PS-*b*-P2VP-*b*-PEO materials were immersed into a deionized water bath at RT for 5 min to produce a macroporous substructure. Importantly, in order to facilitate the peeling of the terpolymer thick films from their substrates, the silicon pieces were treated by an oxygen plasma in a home-made chamber (plasma conditions: 45W, 75 mTorr O_2 , 10min) prior to their use. Such a plasma treatment makes that the blended PS-*b*-P2VP-*b*-PEO thick films are detached from their substrate after several minutes spent in the water bath.

Fabrication of nanostructured asymmetric films and monoliths by SVA: Prior to placing the blended PS-*b*-P2VP-*b*-PEO material in the SVA chamber, the polymeric film floating in the non-solvent bath were deposited on a PVDF support, and the excess of water (mainly visible droplets) has been removed by absorbing the liquid with a Kimtech wipe. The self-assembly of PS-*b*-P2VP-*b*-PEO terpolymers was promoted by exposing films for different

times (ranging from 0h to 5h) to a continuous stream of CHCl_3 vapor produced by bubbling nitrogen gas through the liquid solvent as described previously.²⁹ This continuous flow system was used to control the CHCl_3 vapor pressure in the chamber by dilution with a separate N_2 stream so that a solvent vapor consisted of 32 sccm CHCl_3 vapor and 8 sccm N_2 (total 40 sccm). Importantly, the HPL structure produced on the dense top layer during the early stage of the SVA treatment grows at the expense of the sponge-like substructure until the formation of a monolith entirely composed of the network structure.

SEM, AFM and DSC characterizations: Scanning electron microscopy (SEM, Hitachi S-4800) was used at an accelerating voltage of 5 kV to acquire images of both asymmetric thick films and monoliths formed by a blend of PS-*b*-P2VP-*b*-PEO chains and a short PS homopolymer. Atomic force microscopy (AFM Nano-Observer, CS Instruments) was used in tapping mode to characterize the surface morphology of PS-*b*-P2VP-*b*-PEO/hPS thick films. Silicon cantilevers (PPP-NCH, Nanosensors) with a typical tip radius of ~5 nm were used. The resonance frequency of the cantilevers was about 235 kHz. Prior AFM and SEM imaging, blended PS-*b*-P2VP-*b*-PEO/hPS thick films were treated with a fluorine plasma in a home-made chamber to improve the AFM topographic image contrast (plasma conditions: 45 W, 75 mTorr CF_4 , and 90s). The thermal study was conducted using a TA Instruments DSC Q20 calibrated with indium. A sample with a weight of ~5 mg was heated with a rate of $20^\circ\text{C min}^{-1}$ to 370°C to promote the phase-separation of the PS-*b*-P2VP-*b*-PEO chains. This sample was then cooled to -50°C with a rate of $20^\circ\text{C min}^{-1}$, followed by heating again prior to obtain the DSC endotherm.

Water flux performances, bovine serum albumin and 10 nm gold nanoparticle filtrations: The water permeability of the different PS-*b*-P2VP-*b*-PEO/hPS thick films was measured in a 10 mL filtration cell (Amicon 8010 stirred cell) connected to a water reservoir and a compressed air line. The measurements were performed on 2.5cm diameter PS-*b*-P2VP-

b-PEO/hPS material discs supported by a high permeable hydrophilic PVDF material (water permeability of 2960 L h⁻¹ m⁻² bar⁻¹, see **Fig. S8**) at pressures between 0 and 1.5 bar. The mass of water passing through the stacked materials (permeate) was recorded using a connected balance at regular time intervals for 10 min. Water temperature was maintained at 20°C during the measurements while the error bars were calculated from 3 different samples (see **Tables S1-3**). Aqueous solutions containing monodisperse 10 nm gold NPs (Sigma-Aldrich, the commercial solution was diluted 15 times in Milli-Q ultrapure water) or BSA (Sigma-Aldrich, dispersed in Milli-Q ultrapure water to reach 0.5 g/L at pH = 7) were used to evaluate the rejection of the NIPS and NIPS-SVA made (45 min, CHCl₃) PS-*b*-P2VP-*b*-PEO/hPS thick films. Filtrations were performed at 1 bar with the filtration setup used to determine the water permeability. An UV-visible spectrometer (Shimadzu, UV-2401 PC) was used to measure the 10 nm gold NP concentrations in the feed, permeate and retentate solutions by recording its UV-visible spectrum over the 400-800 nm spectral range. The BSA concentrations of both permeate and retentate, were measured at 278 nm with the same UV-visible equipment.

Acknowledgements

This work was performed within the support of the ANR JCJC AFM_Ring project, grant ANR-18-CE09-00xx of the French Agence Nationale de la Recherche and through the project TAKCLE/AXE-Membrane du Futur/09-2020 founded by the Institut Européen des Membranes (IEM).

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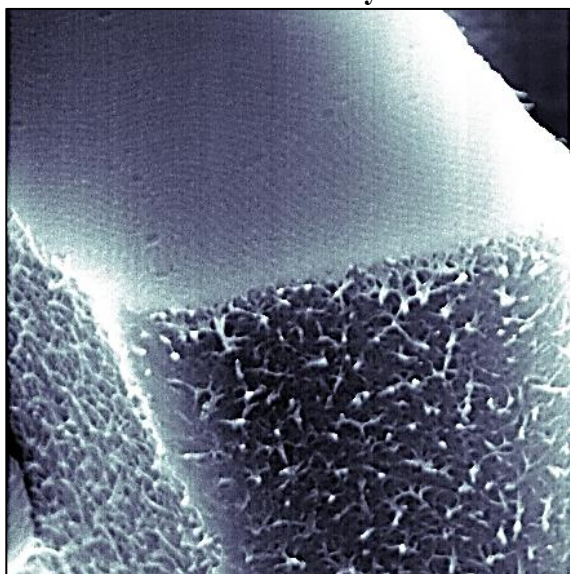
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Supporting Information

Asymmetric Solvent-Annealed Triblock Terpolymer Thick Films Topped by a Hexagonal Perforated Lamellar Nanostructure

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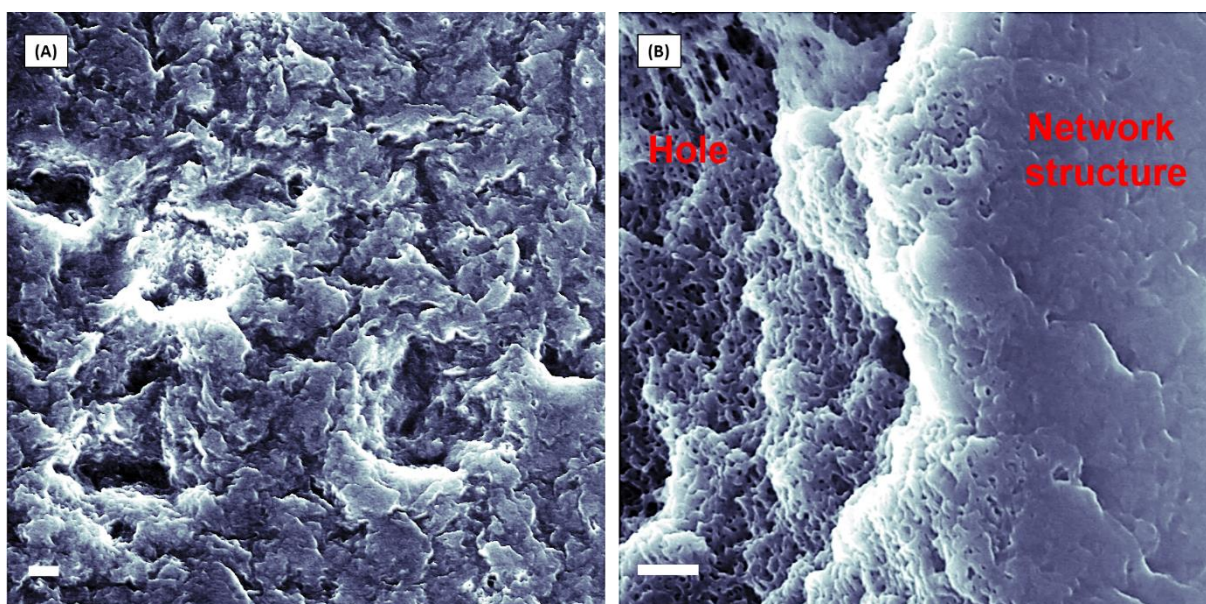


Figure S1: (a) ($10 \times 10 \mu\text{m}$) and (b) ($5 \times 5 \mu\text{m}$) top view SEM images taken from the bottom surface of a PS-*b*-P2VP-*b*-PEO/hPS thick film generated by NIPS. A rough bottom surface, consisting of big holes/cavities and an open network structure, can be observed with no evidence of preferential interactions of the PEO and P2VP domains with the silicon substrate. Scale bars: 500 nm.

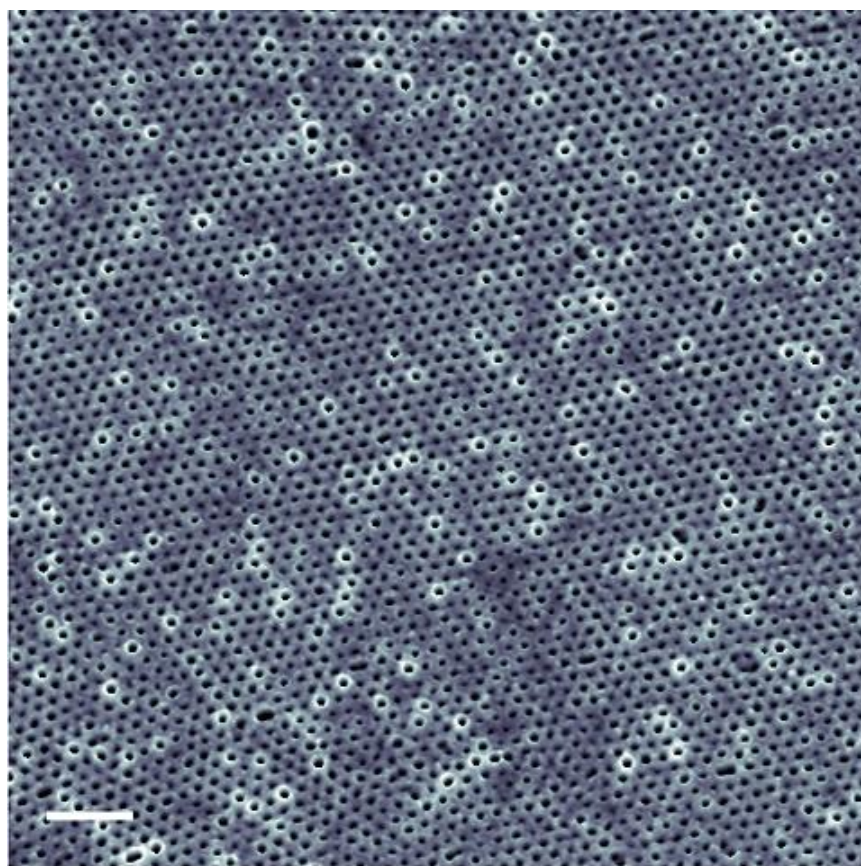


Figure S2: Top view SEM image taken from a corner region of the top surface of a PS-*b*-P2VP-*b*-PEO/hPS thick film generated by NIPS, which shows PEO-based nanopores arranged into a well-ordered hexagonal array with a period of ~ 32.5 nm. Scale bar: 200 nm.

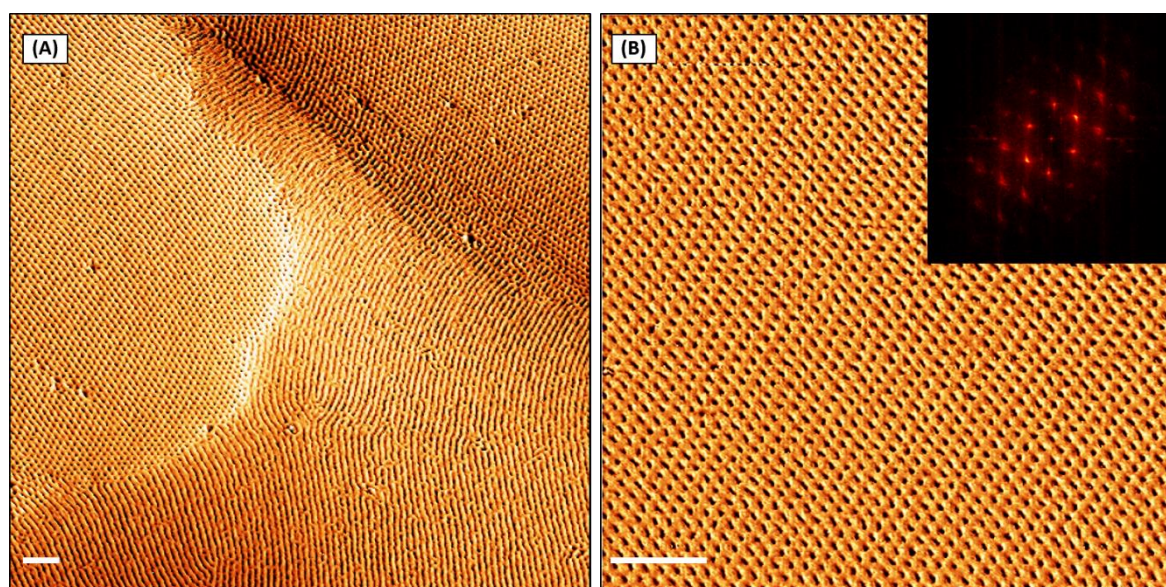


Figure S3: (a) ($4 \times 4 \mu\text{m}$) and (b) ($1.5 \times 1.5 \mu\text{m}$) AFM topographic views showing different plane orientations of the HPL nanostructure formed on the top surface of a solvent-annealed (45 min, CHCl_3) PS-*b*-P2VP-*b*-PEO/hPS thick film. The 2D-FFT indicates that the PS (bright) perforations are arranged into a rhombic array with the lattice parameters $a = 38$ nm and $b = 29$ nm along HPL (110) plane. Scale bars: 250 nm.

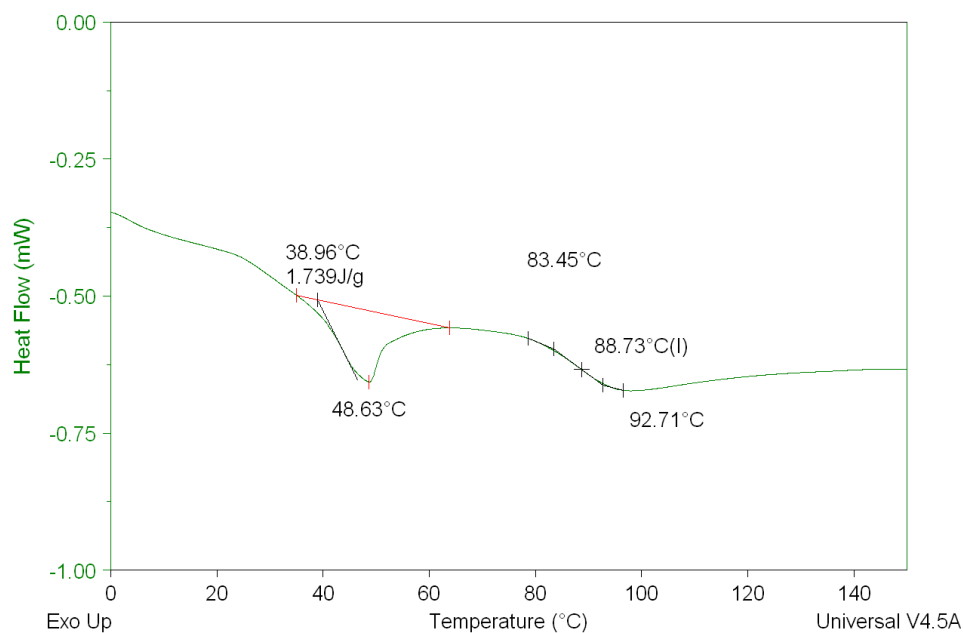


Figure S4: DSC trace of a PS-*b*-PS-*b*-PEO triblock terpolymer powder that was heated with a rate of 20°C min⁻¹ to 370°C, followed by cooling to -50°C with a rate of 20°C min⁻¹, and then by heating (rate of 20°C min⁻¹) again to obtain the DSC endotherm.

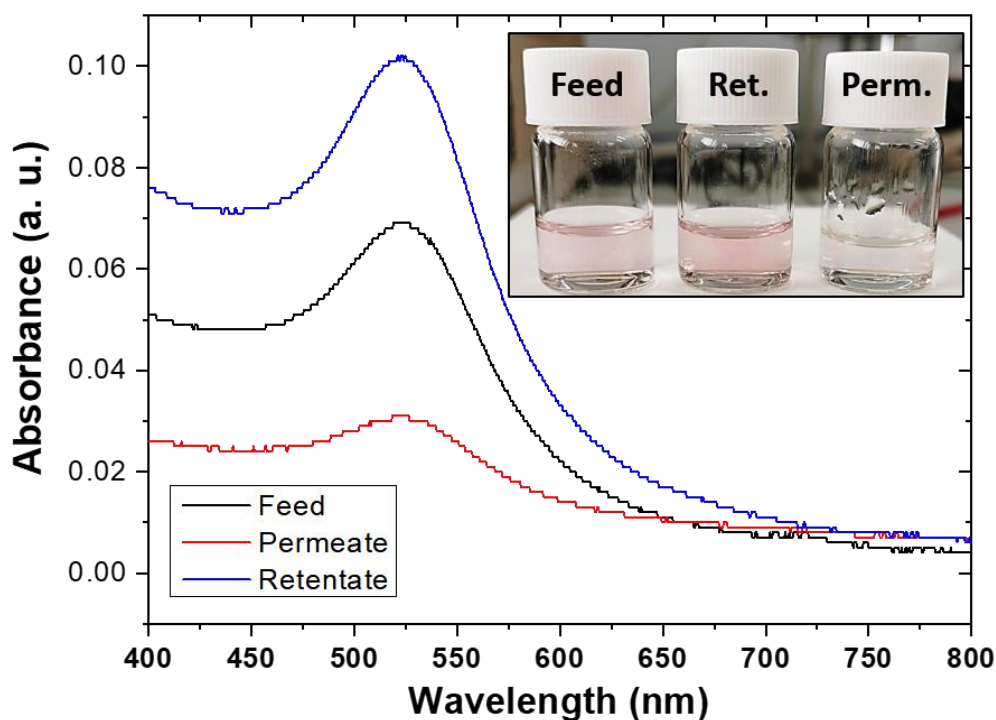


Figure S5: UV-visible spectra of the feed, retentate and permeate recorded over the 400-800 nm spectral range. All the UV-visible spectra mainly consists of a peak centered at approximately 520 nm that corresponds to the surface plasmon polariton generated by the 10 nm gold NPs. Inset: Photograph showing the colored (left) feed and (middle) retention solutions as well as the (right) nearly colorless permeate.

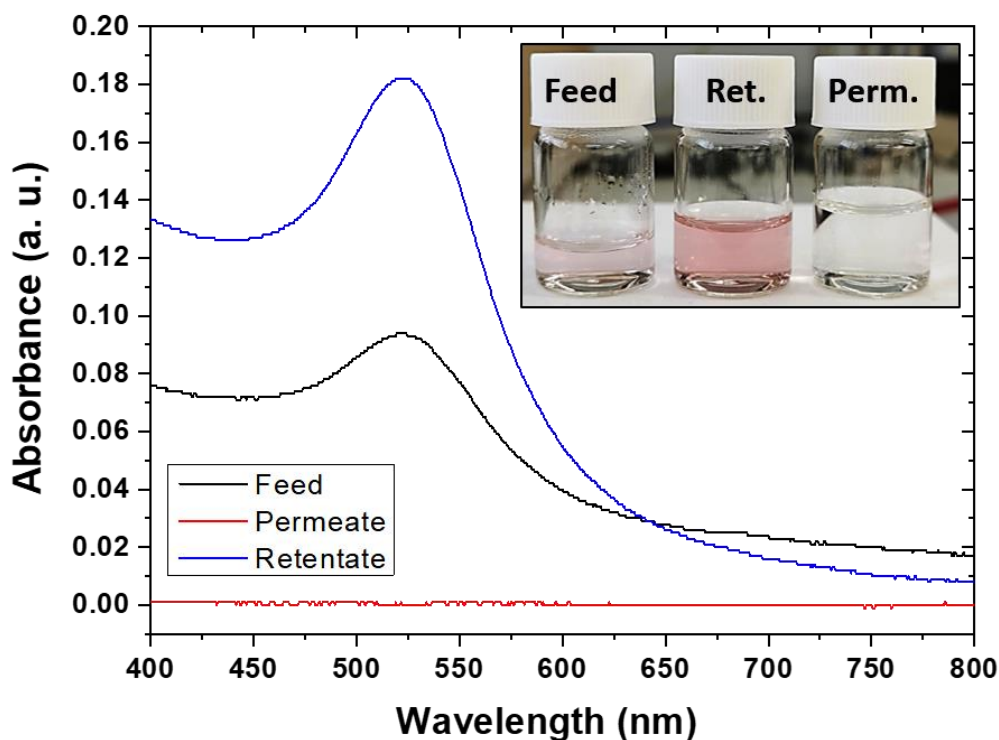


Figure S6: UV-visible spectra of the feed, retentate and permeate recorded over the 400-800 nm spectral range. The UV-visible spectra of the colored feed (black curve) and retention (blue curve) solutions mainly consists of a peak centered at approximately 520 nm that corresponds to the surface plasmon polariton generated by the 10 nm gold NPs. Conversely, no absorption peak is observed at 520 nm on the UV-visible spectrum (red curve) of the colorless permeate, which indicates an efficient rejection of the 10 nm gold NPs. Inset: Photograph showing the colored (left) feed and (middle) retention solutions as well as a (right) colorless permeate.

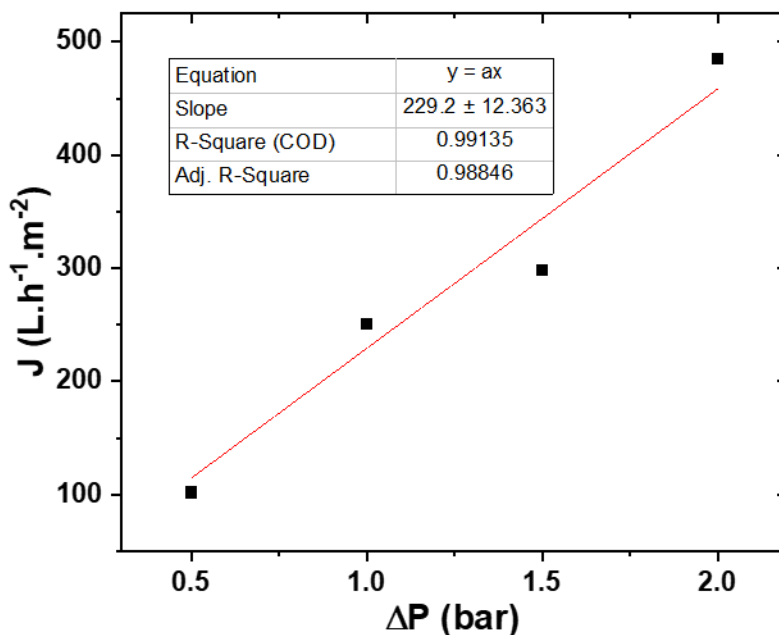


Figure S7: Water flux, J , of a solvent-annealed (2h, CHCl_3) PS-*b*-P2VP-*b*-PEO/hPS thick film exhibiting a water permeability of $\sim 230 \text{ L h}^{-1} \text{ m}^{-2} \text{ bar}^{-1}$.

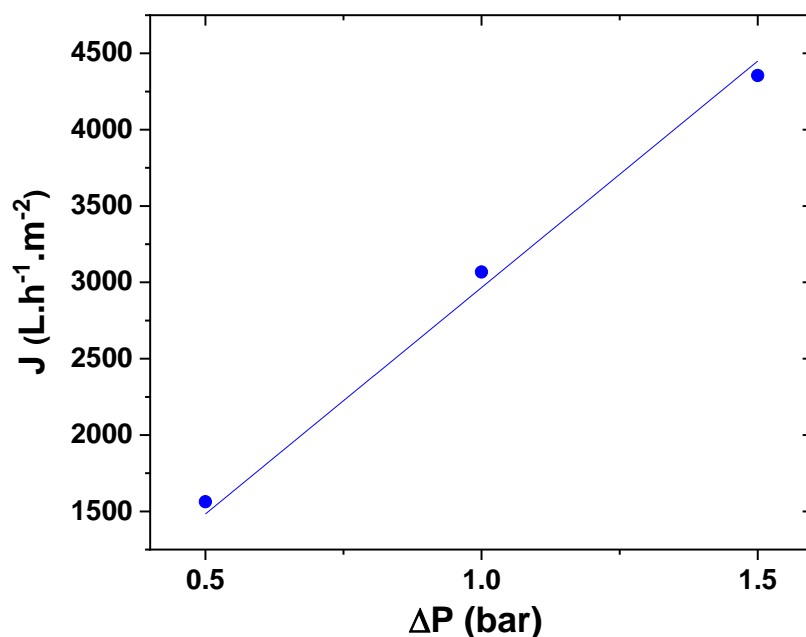


Figure S8: Water flux, J , of the hydrophilic PVDF support used in this work. The commercial PVDF support exhibits a water permeability of $2960 \text{ L h}^{-1} \text{ m}^{-2} \text{ bar}^{-1}$.

Pressure (Bar)	Sample 1 ($\text{L.m}^{-2}.\text{h}^{-1}$)	Sample 2 ($\text{L.m}^{-2}.\text{h}^{-1}$)	Sample 3 ($\text{L.m}^{-2}.\text{h}^{-1}$)	Mean Value ($\text{L.m}^{-2}.\text{h}^{-1}$)	SD Value ($\text{L.m}^{-2}.\text{h}^{-1}$)
0.5	220	138	97	152	63
1.0	441	280	200	307	123
1.5	690	458	379	509	162

Table S1: Permeability test results for three different NIPS-made PS-*b*-P2VP-*b*-PEO/hPS thick film samples measured at pressures between 0 and 1.5 bar. Mean and standard deviation (SD) values of the each permeability result are also given.

Pressure (Bar)	Sample 1 ($\text{L.m}^{-2}.\text{h}^{-1}$)	Sample 2 ($\text{L.m}^{-2}.\text{h}^{-1}$)	Sample 3 ($\text{L.m}^{-2}.\text{h}^{-1}$)	Mean Value ($\text{L.m}^{-2}.\text{h}^{-1}$)	SD Value ($\text{L.m}^{-2}.\text{h}^{-1}$)
0.5	282	417	323	341	69
1.0	850	925	692	822	119
1.5	1482	1433	1141	1352	184

Table S2: Permeability test results for three different solvent-annealed (45 min, CHCl_3) PS-*b*-P2VP-*b*-PEO/hPS thick film samples measured at pressures between 0 and 1.5 bar. Mean and standard deviation (SD) values of the each permeability result are also given.

Pressure (Bar)	Sample 1 (L.m⁻².h⁻¹)	Sample 2 (L.m⁻².h⁻¹)	Sample 3 (L.m⁻².h⁻¹)	Mean Value (L.m⁻².h⁻¹)	SD Value (L.m⁻².h⁻¹)
0.5	13	8	25	15	9
1.0	22	22	43	29	12
1.5	42	37	58	46	11

Table S3: Permeability test results for three different solvent-annealed (5h, CHCl₃) PS-*b*-P2VP-*b*-PEO/hPS thick film samples measured at pressures between 0 and 1.5 bar. Mean and standard deviation (SD) values of the each permeability result are also given.