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High-pressure A-site manganites: structures and magnetic properties.

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Abstract:

Compounds of general formula ABO_3 are widely studied due to their versatility regarding chemical composition, structure and physical properties. The use of high-pressure and high-temperature synthesis method allows the introduction of small Mn^{2+} cation into the often-large *A* site, giving rise to the so-called *A*-site manganites, with a few exceptions to this rule. Polymorphism, structural features and magnetic properties of A-site manganites are revisited in this short review dedicated to Miguel A. Alario y Franco on his 80th birthday. We extend this dedication to John B. Goodenough and J. Paul Attfield on their 100th and 60th birthdays in recognition of their ground work. Happy birthday all three!

Introduction:

 ABO_3 oxides are of wide interest in the search for new functional materials. Their chemical and structural versatility provides many possibilities in the design of materials with new or enhanced physical properties, which motivates a continuous production of new phases. A broad range of chemical compositions ^[1,2] and polymorphs are accessible by controlling the relative ionic radii of *A* and *B*-site cations, as defined by the Goldschmidt tolerance factor (t = [(r₀+r_A)/ $\sqrt{2}(r_0+r_B)$]), ^[3] and the synthesis conditions. In particular, high-pressure and high-temperature (HPHT) synthesis method, which allows the stabilisation of metastable phases ruled by kinetics and hosting cations in unusual oxidation states or coordination environments, has lately raised interest in the growing family of the so-called *A*-site manganites as new functional materials.

A-site manganites are ABO_3 compounds where the A sites are occupied by Mn (generally 2+) cations. Partial substitution of A and/or B-site cations induces ordered derivatives with increased chemical, structural and functional complexity. The fast appearance of new phases with properties in the fields of magnetism, spintronics and multiferroics motivates this short review, where we aim to establish the relation between the chemical composition, structural details and magnetic properties of the known A-site manganites.

The article is organised in four different sections. The first two classify the high-pressure A-site manganites into their two known polymorphic families: perovskite- and corundum-related structures, with their different superstructures considering their A/B cation order. The third section briefly mentions some structural aspects to reconsider. Finally, we propose new directions in the design of novel materials.

A) <u>Perovskite-related A-site manganites.</u>

Cation order into the parent MnBO₃ "simple" perovskite structure (Pv) can be induced into A and/or B sublattices with 1:1, 1:2 and 1:3 order ratios, giving rise to the observed $AMnB_2O_6$ A-site ordered or $Mn_2BB'O_6$ B-site ordered double perovskites (DPv), $AMn_{0.5}Mn'_{0.5}BB'O_6$ double double perovskites (DDv), $Mn_3MnB_2O_9$ triple perovskites (TPv) and $AMn_3B_4O_{12}$ quadruple perovskites (QPv). All known examples are summarised in the following sections, with a single case for simple Pv, one A- and twenty-one B-site ordered DPv, seventeen DDPv, two TPv and many QPv. The discussed structures are shown in Figure 1, where the occupational, rotational and displacive distortion modes relating each superstructure to the parent Pv structure are identified.

Complex magnetic properties are observed throughout the whole family, with coherent evolution of transition temperatures and magnetic moments within each superstructure type in terms of the number of unpaired electrons for *B*-site cations. Tables 1 and 2 summarise the main structural and magnetic features for *B*-site DPv and DDPv respectively.

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Figure 1. Perovskite related structures of *A*-site manganites. Cubic parent perovskite (Pv) in the middle, 1:1 cation orders at the *A* (DPv_AMn_{0.5}Mn'_{0.5}B₂O₆), *B* (DPv_Mn₂BB'O₆), *A* and *B* sites (DDPv_AMn_{0.5}Mn'_{0.5}BB'O₆); 1:2 order of *B* sublattice (TPv_Mn₃MnB'₂O₉) and 1:3 order of *A* sites (QPv_AMn₃B₄O₁₂). Colours in general compositions correspond to those of octahedra or spheres in the structures. *A*-site 4-fold coordination polyhedra are included below each superstructure for comparison. The labels in green, red, and grey relate the observed symmetry at room temperature to the parent Pv from occupational, rotational and displacive modes respectively, using a parent Pv with *A* sites at ($\frac{1}{2}$ $\frac{1}{2}$). Space groups in brackets show ideal symmetry for each superstructure attending only at the cation order.

1) Simple:

MnVO₃ is the unique simple *A*-site manganite that can be recovered from HPHT to ambient conditions. It crystallises with the *Pnma* Pv structure in a $\sqrt{2a_p} \times 2a_p \times \sqrt{2a_p}$ supercell, where $a_p \sim 3.8$ Å is the cell parameter of the cubic Pv. It shows spintronic properties arising from the combination of localised Mn²⁺ 3*d*⁵ and itinerant V⁴⁺ 3*d*¹ electrons. ^[4] It is metallic between 2 and 300 K, while it shows a complex magnetic behaviour at low temperatures. Below $T_N = 50$ K, it develops an incommensurate magnetic structure with [*k_x* 0 0] propagation vector and either a helical spin structure or a spin density wave of collinear moments with modulated amplitude (as shown in Figure 2a). Magnetic frustration factor $f = |\theta|/T_N \sim 3$ sets the starting point for the complex magnetism of *A*-site manganites.

Unquenchable high-pressure A-site simple perovskites have also been observed for $MnSnO_3$ and $MnTiO_3$ oxides. ^[5] The three simple A-site perovskites show lower pressure corundum-related polymorphs which will be described in later sections.

2) Double:

Double perovskites are stabilised by 1:1 cation order into either *A* or *B* sites. Again, there is a single example of $AMnB_2O_6$ DPv *A*-site manganite: CaMnTi₂O₆. ^[6] It crystallises with tetragonal symmetry in a $2a_p \times 2a_p \times 2a_p$ supercell (S.G. $P4_2mc$) and Ca and Mn order into columns along the *c* axis. Mn sites show alternated tetrahedral and square-planar coordination along its column, as shown in Figure 1. The cooperative off-centre of Ti in octahedral coordination via second-order Jahn-Teller and Mn shifts in the square planes, induce a ferroelectric – paraelectric transition at $T_c = 630$ K. Above this temperature, an order-disorder phase transition results in the non-isomorphic supergroup $P4_2/nmc$ of the closely related *A*-site ordered DPv CaFeTi₂O₆. ^[7] This makes CaMnTi₂O₆ the first ferroelectric compound with *A*-site columnar ordered DPv structure. It is also multiferroic, with a weak spin-polarisation coupling ^[8] below its antiferromagnetic transition at $T_N = 10$ K. The low temperature magnetic phase ($P4_2mc$) is predicted as a *C*-type AFM with $k = [0 \ 0 \ 0]$ propagation vector and collinear alignment of the spins along the *c*-axis, *i.e.* parallel to the columnar order and perpendicular to the square-planes. ^[9] New improper ferroelectrics with different degrees of cation order can be accessed for instance in Ca_{2-x}Mn_xTi₂O₆. ^[10]

Among *B*-site ordered Mn₂*BB*'O₆ *P*₂₁/*n* double perovskites with $\sqrt{2a_p} \times \sqrt{2a_p} \times 2a_p$ supercell, a simple classification can be made considering the relative oxidation states of *B/B*' cations, *i.e.* 3+/5+, 2+/6+ and +/7+. In all cases, the electronic configuration of *B*-site cations plays a key role in their magnetic properties, for which they can be further classified into dⁿ/d⁰(d¹⁰) and dⁿ/dⁿ materials. Two families of compounds with *B*'= Sb⁵⁺, Te⁶⁺ are reported in the first case (Mn₂ScSbO₆, Mn₂CrSbO₆, Mn₂CoTeO₆ and Mn₂MnTeO₆), all showing simple magnetic

behaviours with collinear antiferromagnetic (AFM) structures with $k = [0 \ 0 \ 0]$ or [½ 0 ½] propagation vectors (Figure 2b-top). ^[11,12,13,14] The only exception is Mn₂FeSbO₆, where competing Mn-O-Mn and Mn-O-Fe $d^6 - d^6$ magnetic interactions frustrate the system into a complex helical spin structure with propagation vector $k = [0 \ k_y \ 0]$ (shown in Figure 2b-bottom). ^[15] The combination of paramagnetic cations into both *B/B*' sites is only reported for the "all transition metal" *B*' = Re family. Strong magnetic interactions make Mn₂FeReO₆ a high temperature ferromagnet (FM) with $T_C = 520$ K, and a colossal magnetorresistance switch induced by a low temperature spin reorientation transition. ^[16,17] All other Mn₂BReO₆ belong to a complex series of frustrated magnets with B^{2+} and Re⁶⁺ perpendicular sublattices (Figure 2b-bottom right), ^[18,19,20] except for Mn₂LiReO₆ DPv. This is the only case among all *A*-site manganites with B^+/Re^{7+} and a giant magnetic coercivity due to magnetocrystalline anisotropy, enhanced by symmetry restrictions and without spin-orbit coupling. ^[21] A special case is Mn₂(Fe_{0.8}Mo_{0.2})MoO₆,^[22] which was discovered as a line phase from a compositional modulation motivated from the parent Mn₂FeMO₆ corundum-related phase. It exhibits a Fe-Mo ferrimagnetic (FiM) transition at $T_C = 194$ K and a Mn AFM at $T_N = 45$ K. It demonstrates that both polymorphs can be tuned by slight variation of the composition.

 $MnRMnB'O_6$ (where *R* stands for rare earth) DPv structures with Mn^{2+}/R^{3+} disordered over the *A* sites and Mn^{2+}/B^{5+} ordered with a rock-salt pattern over the *B* sites are reported for B' = Sb and Ta when R^{3+} cations are not large enough to induce further order into the *A* sublattice (*vide infra*). ^[23,24,25] This evidences the polymorphism throughout two series of compounds, suggesting chemical pressure as a key factor on structural control. In a similar case, mixed *A* sites are also reported for CaMnMnWO₆. ^[24] As summarised in Table 1, the independent magnetic transitions are observed for Mn²⁺ and R³⁺ and a unique case of spin glass behaviour among all DPv *A*-site manganites for CaMnMnWO₆, due to *B*-site geometric frustration.

Table 1. Main magnetic features of Mn₂BB'O₆ DPv with B/B' rock-salt order. S.G. P2₁/n, cell parameters $\sqrt{2a_p}$ x $\sqrt{2a_p}$ x $2a_p$. Horizontal lines separate materials with $d^{10}(d^0)$ B' cations and d^n/d^n compounds at the top part. The lower half shows isostructural DPvs with mixed B or A occupancies.

В	B'	d _B		d _{B'}	<i>Т</i> м/ <i>Тс</i> (K)	f	K 1	Other properties	Ref
		t _{2g}	eg	•		0			
Sc	Sb	0	0	10	22	4.3	[000]	Polymorphism	11
Co	Те	5	2	10	27		[000]	Polymorphism	14
Mn	Те	3	2	10	36	5.1	[1/2 0 1/2]	Polymorphism	20
Cr	Sb	3	0	10	55	4.0	[1/2 0 1/2]	Polymorphism	12
Fe	Sb	3	2	10	60	3.0	[0 0.426 0]	Elliptical helix	15
Li	Re	0	0	0	109	1.2	[000]	Giant coercivity	21
			$\langle \ \rangle$					Polymorphism	
Ni	Re	6	2	1	80	0.02	[000]	Spin rotation	20
Co	Re	5	2	1	94	3.6	[1/2 1/2 0]	Semiconductor	19
								Spin reorientation	
								T_{N2} = 40 K to k_0	
Mn	Re	3	2	1	109	1.4	[1/2 1/2 0]	T _{Re} = 99 K	18
Fe	Re	3	2	2	520	0.9		T _{Mn} = 250 K	16
								T _{C3} = 75 K: spin	17
								reorientation and	
								Magnetoresistance	
$Fe_{0.8}Mo_{0.2}$	Мо	5	5/1	1	194	0.3	[000]	T _{N2} = 45 K	22
A/A'	B/B'	e⁻A	dB	d _{B'}	T_N/T_C (K)	f	k 1	Other properties	Ref
MnEu	MnSb	d ⁵ /f ⁶	5	10	61			T _{N2} = 38 K	23
MnGd	MnSb	d ⁵ /f ⁷	5	10	67.5	0.5			23
CaMn	MnW	s ⁰ /d ⁵	5	10	8	18		Spin glass	24
								Polymorphism	
SmMn	MnTa	f ⁵ /d ⁵	5	10	35	4.5		Polymorphism	24
									25
EuMn	MnTa	f ⁶ /d ⁵	5	10	26			T _{C2} = 13 K	25
GdMn	MnTa	f ⁷ /d ⁵	5	10	45	1.3		T _{C2} = 20 K	25
YMn	MnTa	d ⁰ /d ⁵	5	10	28	14			25
LuMn	MnTa	f ¹⁴ /d ⁵	5	10					25
Mn	Mn _{0.5} Co _{0.5} /Te	5	5/7	10	31		[1/2 0 1/2]	Magnetic phase	14
					23		[000]	coexistence	
$Mn_{1.5}Co_{0.5}$	Co/Te	5/7	7	10	25		[000]	Polymorphism	14

* Missing values imply the compound is not fully studied.



Figure 2. Representative examples of all types of reported magnetic structures for Pv (a), DPv (b), DDPv (c) and TPv (d) A-site manganites. e) shows the magnetic structures of Mn_2O_3 QPv. Figures adapted from references [4, 14, 15, 18, 31, 32, 34, 37, 53].

3) Double-double

Cation order into both A and B sites of the perovskite structure can happen into columnar (C), layered (L) or rock-salt (RS) arrangements. The combination of these three different patterns into both sublattices simultaneously allows 9 different types of doubly ordered perovskites. [2] Among them, only two combinations have ever been reported: layered/columnar patterns at A/A' sites with rock-salt order of B/B' cations. Layered and rocksalt orders are the most stable cation arrangements for A and B sublattices respectively and consequently there are many doubly ordered perovskites combining L_A + RS_B. [26,27,28,29,30] However, such order has not been observed with Mn in the layered position. The likely reason for this is the stress induced into the oxygen lattice due to Mn insertion. Nevertheless, the columnar order seems to favour small cations and thus it is much less stable among the usually large A-site cations, which makes it much rarer. Therefore, the amount of C A DPv structures is scarce in comparison with L A DPv. For the same reason, among doubly ordered perovskites, the C A + RS B combination had never been observed before the first A-site manganites with the so-called double double perovskite structure (DDPv), reported back in 2016. [23] As it happens in CaMnTi₂O₆ and CaFeTi₂O₆ (the only examples of C_A DPv up to date), $^{[6,7]}$ the combination of large (Ca²⁺ or R^{3+}) with the small (Fe²⁺ or Mn²⁺) cations proved the major importance of having sufficiently different ionic radii over charge mismatch in the stabilisation of columnar cation order within the A sites, ita vero Ca and Mn are isovalent in this structure. The DDPv structure is notable for having 5 independent cation sites with 4 different coordination environments. As described in the previous section, Mn columns alternate tetrahedral and square-planar coordinations, revealing a close structural relation between these structures, with just an additional R¹⁺ B-site ordering mode but with the same A-site and tilt M⁴⁺ and R⁴⁺ modes respectively as shown in Figure 1.

The DDPv structure has demonstrated a wide chemical and physical versatility in the last few years, where different cations could be stabilised within all 5 cation sites. This chemical flexibility is intrinsically coupled to a broad variety of magnetic properties, so compounds with ferro-, ferri-^[31] and antiferromagnetic behaviours have been reported with magnetic transitions at 42 K < T_C < 560 K, including strong magnetic anisotropy and crystal field effects for Mn*R*MnSbO₆, ^[32] spin reorientation transitions in MnNdMn*B*'O₆ (*B*' = Sb, Ta) ^[23,25] and CaMnFeSbO₆, ^[33] magnetoresistance as in CaMn_{0.5}Cu_{0.5}MnReO₆ and Ca_xMn_{2-x}FeReO₆ solid solution for 0.17 < x < 1.73 ^[34,35], and a cluster spin glass behaviour for CaMnFeTaO₆. ^[36] All reported DDPv show simple magnetic structures with propagation vector *k* = [0 0 0]. However, different observed spin arrangements can be summarised into antiparallel sublattices along the *c* or the *a* axis (Figure 2c-top) and parallel or perpendicular FM sublattices (Figure 2c-bottom). Order-disorder phase transitions are reported at high temperature in for instance CaMnMnWO₆ and *R*MnMnTaO₆, resulting in the already described DPv. ^[24,25] The main features of this family are condensed in Table 2.

compa	irison. Propaga	ation vecto	r in all s	studiec	cases	$s K_1 =$	[0 0 0].			
Α	В	B'	s/f _A		d _B	d _{B'}	T _N /T _C (K)	f	Other properties	Ref
				t_{2g}	e_g					
La	Mn	Sb	0	3	2	10	48			23, 32
Pr	Mn	Sb	2	3	2	10	89		T _{N2} = 75 K.	23, 32
									Magnetic anisotropy	
Nd	Mn	Sb	3	3	2	10	76	1.1	T _{N2} = 42 K.	23, 32
									Spin reorientation,	
									Magnetic anisotropy.	
Sm	Mn	Sb	5	3	2	10	80	1.6	T _{N2} = 39 K	23
La	Mn	Та	0	3	2	10	49	3.6		25
Nd	Mn	Та	3	3	2	10	65	2.2	T _{C2} = 40 K	25
									Spin reorientation,	
									Magnetic anisotropy	
Sm	Mn	Та	5	3	2	10	69	3.1	T _{C2} = 18 K	24,25
Ca	Cr	Sb	0	3	0	10	49	2.4		33
Ca	Fe	Sb	0	3	2	10	55	0.65	T _{C2} = 21 K,	33
									Spin reorientation	
Ca	Fe	Та	0	3	2	10	51	1.4	T _{C2} = 20 K,	36
									Cluster spin glass	
Ca	Mn	W	0	3	2	10	45	4.4	Polymorphism	24
Ca	Ni	Re	0	6	2	1	152	0.99	Ferromagnetic	31
Ca	Co	Re	0	5	2	1	188	1	Ferrimagnetic	31
Ca	Mn	Re	0	3	2	1	120		T _{A'} = 100 K	34
Ca	Fe	Re	0	3	2	2	500		T _{A'} = 70 K	34
Α	A'	B/B'	SA	d _{A'}	dB	d _{B'}	T _N ∕T _C (K)	f	Other properties	Ref
Ca	$Mn_{0.5}Cu_{0.5}$	Fe/Re	0	5/9	5	2	560		T _{A'} = 160 K	34
									Magnetoresistance	
Ca	Cu	Fe/Re	0	9	5	2	567	0.9	Butterfly-like	a)
									magnetoresistance	

Table 2. Main magnetic features of $AMnBB'O_6$ DDPv with A/Mn columnar and B/B' rock-salt orders. S.G. $P4_2/n$, cell parameters $2a_p \times 2a_p \times 2a_p$. Horizontal lines separate materials with $A = R / Ca f^n / s^0$ cations and B/B' 3+/5+ vs 2+/6+ oxidation states. The bottom part shows mixed A' cations and the only non-Mn based DDPv for comparison. Propagation vector in all studied cases $k_1 = [0, 0, 0]$

* Missing values imply the compound is not fully studied. ^{a)} Submitted.

4) Triple

A-site manganites with the triple perovskite structure have only been reported in 2021.^[37,38] Mn₃MnB'₂O₉ with $B' = Nb^{5+}$ and Ta⁵⁺ are obtained from high pressure and high temperature phase transformation of the Mn₄B'₂O₉ room pressure polymorphs, stabilising a 2:1 cation order among the *B* sites. This structure adopts a structural distortion from the archetype TPv Ba₃SrTa₂O₉ (*S.G. P-3m*1), where the combination of Λ_1 (2:1 cation order), R₄⁺ (octahedral tilt) and Γ_4^- modes (displacive) results in independent crystallographic sites for all three *A* and *B* cations into *Cc* space group with a $\sqrt{6a_p} \times \sqrt{2a_p} \times 2\sqrt{3a_p}$ supercell. According to GKA rules, ^[39] the expected AFM interactions cannot be satisfied for the Mn²⁺ spins at the *B*-site. Thus, a complex magnetic behaviour arises with three subsequent transitions on cooling. Below the first transition (*T_N*), these compounds adopt a collinear AFM structure with spins along the *c* axis and propagation vector $k = [0 \ 0 \ 0]$. The frustration of the *B* sites induces a modulation of the magnetic moments below *T_M* into a complex spin density wave with $k = [k_x \ 0 \ k_z]$. Both incommensurate components evolve continuously down to *T_L*, where they lock into a $k = [1/3 \ 0 -1/6]$. The value of

this propagation vector is dictated from the 2:1 *B*-site ordering and *A*-*B* frustrated interactions. A schematic representation of the complex SDW magnetic structure at low temperature is shown in Figure 2d. A detailed study on the frustrated magnetic interactions is included in the electronic supplementary information file for reference [37]. The room pressure multiferroic polymorphs are also recoverable under heating from the HPHT phase. The UV-Vis data shows a 25 % band gap reduction with respect to the $Mn_4B'_2O_9$ room pressure polymorphs due to the different connectivity of the $B'O_6$ octahedra in both structures.

5) Quadruple

Quadruple perovskites are an extended family of oxides [40,41,42] where AA'₃B₄O₁₂ exhibit 1:3 order of A and A' cations, which may also coexist with 1:1 order between B and B' sites into AA'₃B₂B'₂O₁₂ QDPv. ^[43] They mainly crystallise in the Im-3 cubic space group with a $2a_p \times 2a_p \times 2a_p$ supercell, but they often show lower symmetry due to charge-orbital orders. QPv A-site manganites are known with Mn³⁺ sitting at the square-planar A' sites only but also with aliovalent Mn occupying other cation positions. Among them, A⁺Mn³⁺₃Mn³⁺₂Mn⁴⁺₂O₁₂ (A = Na),^[44] $A^{2*}Mn^{3*}_{3}Mn^{3*}_{3}Mn^{4*}O_{12}$ (A = Ca, Sr, Cd, Hg, Pb, Mn) [45,46,47,48,49] and $A^{3*}Mn^{3*}_{3}Mn^{3*}_{4}O_{12}$ (A = R, Bi) [50,51,52] compositions are reported. Note only Mn²⁺Mn³⁺₃Mn³⁺₃Mn⁴⁺O₁₂ includes Mn²⁺ at the highly coordinated A site, while A' sites are occupied by Mn3+ in all cases. QPvs are thus an exception to the usual 2+ valence of Mn in A-site manganites. Note also that CaMn7O12 can be obtained at room pressure conditions, resulting as well in an exception to the title compounds. The combination of mixed valence Mn into different crystallographic sites commonly induces complex magneto-orbital textures in these compounds, where ζ -Mn₂O₃ is notable as an "all Mn QPv" with a distortion down to triclinic symmetry. The more common cubic symmetry with Im-3 space group is shown in Figure 1 and the only-due-to-cation-ordering ideal space group Im-3m in brackets. ζ -Mn₂O₃ shows two intricate AFM structures below $T_N = 100$ K, combining a spin cycloid with a helical component (Figure 2e) which develops a ferroelectric polarisation. All of these occurs thanks to the combination of octahedral tilting, Jahn-Teller distortion and charge-orbital-spin ordering. [49,53]

The structural relation between Pv, *A* or *B* site ordered DPvs and DDPv is straightforward, given the related distortion modes, order ratios, tilt systems and coordination environments as described above and summarised in Figure 1. The different order ratio into the 1:2 *B*-site ordered TPvs is easily rationalised through the lambda mode relating the cubic Pv structure with the archetype TPv structure and further similar tilts and displacive modes. However, although the structural relation is undeniable, the 1:3 cation order into the QPv structure is not trivial to compare to 1:1 *A*-site ordering, even though some arguments were eventually reported. ^[54] The presence of a single crystallographic site for both *A* and *A'* sites with large cuboctahedral and square-planar coordinations respectively, in different stoichiometric proportions, requires the use of the M₄⁺ mode to describe the cation order but results in *Im-3m* symmetry in the double supercell. Additional in-phase tilt mode M₃⁺, which also contrasts with the out-of-phase tilt R₄⁺ observed in all other Pv-type *A*-site manganites, sets a further difference for QPvs. This means, M₄⁺ mode is a second order parameter in the case of QPv where M₃⁺ mode drives the 1:3 cation site splitting through octahedral tilt. Note the different symmetry relations in the *A*-DPv and DDPv structures yield to alternating square-planar and tetrahedral coordinations at the *A'* sites, in any case described as 2:1:1 order of *A:A':A''* sites, but not as 3:1 *A:A'* as in the QPv structure.

B) Corundum-related A-site manganites.

As a rule of thumb, when the Goldschmidt tolerance factor t < 0.75 the corundum polymorph will be favoured. This just means that A and B sites have a similar size and it is the reason why many A₂O₃ compounds crystallise in this structure.

In parallel to the previous analysis regarding cation arrangements giving rise to ordered perovskite structures, cation order into the parent A_2O_3 disordered corundum structure (C) has been observed among *A*-site manganites into 1:1 (MnBO₃) and 1:2 (known as Mn₄B₂O₉ for the room pressure polymorph, but effectively Mn₃(MnB₂)O₉) ratios. Further order into doubly ordered MnA'BB'O₆ with 1:1 order into both *A* and *B* sublattices is also common when the chemical composition includes cations with large charge / size mismatches.

Regardless the degree of cation order into each sublattice, A-site Mn corundum related 1:1 ABO₃ compounds are usually referred to as Mn₂BB'O₆. Among them, compounds with B/B' cations similar in charge and ionic radii will result in disordered alternating (00I) or (012) A/B layers into ilmenite (IL, R-3, Γ_2^+ occupational mode) ^[55] and LiNbO₃-type (LN, R3c, Γ_2^{-} occupational mode) [^{56]} structures respectively. Another ordering is possible through the Γ_1^+ occupational mode. It results in the hypothetical R32 structure with ordering in dimers, but this has not yet been observed. Note that $Mn_4B_2O_9$ structure is the closest approach to such R32 structure, with Λ_2 occupational mode allowing a 2:1 Mn: B order that modifies octahedral connectivity every three (00I) layers. This -hhc- sequence turns this structure into a bridge between hexagonal and cubic packing and thus between corundum- and perovskiterelated phases via hexagonal perovskites. The three structures contain only two independent cation crystallographic sites. However, when charge and size difference between B and B' cations is larger, further order can be induced into the B sublattice of an ordered ilmenite (OIL) type structure. In this case, only three different cations are present but, due to symmetry relations, they are distributed over 4 independent crystallographic sites with space group R3. Note OIL shows alternating A/B (00I) layers, while a similar double arrangement into ordered LiNbO₃ (OLN, R3) is also possible in the unique Co₂InSbO₆ ^[57] and not yet observed in A-site manganites. In the same way, partial substitution of Mn can also result in combined cation order into the A sublattice, giving rise to the so-called Ni₃TeO₆-type (NTO, R3) [58] structures, see Figure 3.



Figure 3. Corundum related structures of *A*-site manganites. Disordered parent corundum (C) in the middle, 1:1 cation orders into ilmenite (IL_MnBO_3) and $LiNbO_3$ -type (LN_MnBO_3), 1:2 order into $Mn_4B_2O_9$ and additional order into both sublattices from 1:1 ordered derivatives into $OIL_Mn_2BB'O_6$, $OLN_Mn_2BB'O_6$ and Ni_3TeO_6 -type structure ($NTO_MnA'BB'O_6$). Colours in general compositions correspond to those of octahedra or spheres in the structures. Symmetry relations to the parent C structure are labelled.

A similar charge distribution comparison as that discussed for DPv phases reveals that most corundumrelated Mn₂*BB*'O₆ oxides stabilise 3+/5+ *B*/*B*' cations with IL (*B* = Cr, Fe; *B*' = Sb), ^[12, 59] LN (*B* = Fe; *B*' = Nb, Ta), ^[60] OIL (*B* = In, Fe; *B*' = Sb, Mo) ^[61,62.] and NTO (*B* = Sc, Fe; *B*' = Sb, Mo, W) structures. ^[11,63,64] Only Mn₃WO₆ and the Mn_{3-x}Co_xTeO₆ solid solution with x \ge 1.5 are reported with NTO-type structure and 2+/6+ *B*/*B*' cations. ^[14,65] Also, a single example can be found with +/7+ charge distribution at the *B*/*B*' sites of an OIL-type structure Mn₂LiReO₆. It is the only member of the corundum-related *A*-site manganites with *B*' = Re up to date. ^[21] Due to the stacking of honeycomb and/or triangular magnetic sublattices along the *c* axis, all these compounds are largely frustrated, with complex magnetic phase diagrams including incommensurate helical/conical spin structures, temperature-dependent propagation vectors, lock-in phase transitions and even splitting *k*-vectors into competing ground states. The next subsections include further details on the magnetic properties of known corundum related *A*-site manganites as a function of their *A*:*B* order ratio and are also summarised in Table 3. The magnetic behaviour of these compounds is due to stacking of triangular and honeycomb magnetic sublattices that give rise to strong magnetic frustration; representative examples of the magnetic structures are shown in Figure 4 and discussed in the following subsections.

1) Corundum (no order)

The closely related α -Fe₂O₃ and α -Cr₂O₃ are well-known text book examples of corundum antiferromagnets, ^[66] However, α -Mn₂O₃ adopts a cubic bixbyite-type (C-type R_2O_3 , S.G. *Pbca*) structure. The Jahn-Teller effect of Mn³⁺ cations stabilises the oxygen-deficient fluorite structure rather than the corundum stacking at ambient conditions. ^[67] It is only when using HPHT (12 < P < 17 GPa at 1200 K) that the initially identified as Mn₂O₃ corundum-type appears (ϵ -Mn₂O₃, S.G. *R*-3c). Later on, this phase was properly identified as the charge ordered ϵ -Mn²⁺Mn⁴⁺O₃ ilmenite type structure (S.G. *R*-3) ^[49,68]. Provided the cation disorder from a strict chemical point of view, this phase is still included in this disordered section. Thus, no synthesis conditions have yet been reported to yield the formation of the *R*-3*c* phase in kinetic conditions. Mn₂O₃ shows a rich P-T phase diagram discussed later on.

2) Simple (1:1 A:B order)

 $MnTiO_3$ and $MnSnO_3$ are the only known non-distorted 1:1 ordered corundum derivative A-site manganites with "simple" ABO_3 chemical formula stabilising the two possible cation arrangements (IL- and LN-type structures). At ambient pressure, they crystallise with IL-type structure (I), and under 7 GPa they transform to a LN-type phase (II), ^[69,70] with canted antiferromagnetism below $T_N = 25$ and 50 K respectively and evidence for dielectric properties. A spin-flop transition at ~6 T turns $MnTiO_3$ -I into a magnetoelectric material with proposed ferrotoriodal order. ^[71] The $MnTiO_3$ -II polymorph shows an unusual low field domain reorientation, which gives rise to weak ferromagnetic Dzyaloshinskii Moriya contributions perpendicular to the electric polarisation, thus turning $MnTiO_3$ -II into a multiferroic material. ^[72]

MnVO₃ can be prepared as a distorted ilmenite (S.G. *P*-1) below 4 GPa. ^[73] It presents a weak ferromagnetic transition with T_N = 70 K and θ = -430 K, showing a larger frustration (*f* = 6.1) than its perovskite polymorph (*f* ~ 3) described in section A above.

3) Double (1:1 B:B' order)

As mentioned before, A-site manganites crystallise with IL (R-3) and LN (R3c) type structures for Mn₂BB'O₆ compositions with trivalent first row transition metals at the B sites and pentavalent cations at the B' sites. Despite the charge difference, B/B' cations with similar ionic radii are disordered and only a 1:1 arrangement is observed between Mn²⁺ (A) and B(B') in alternating (00I) or (012) layers respectively. Only four examples have been reported, where Sb⁵⁺ favours the IL order in Mn_2CrSbO_6 and Mn_2FeSbO_6 [^{12,59]} while transition metal Nb⁵⁺ and Ta⁵⁺ cations stabilise the LN structure for Mn₂FeNbO₆ and Mn₂FeTaO₆. ^[60] The different stacking of A and B octahedra arises from the smaller difference in averaged ionic radii for Mn²⁺ (0.83 Å) vs Nb⁵⁺(Ta⁵⁺)/Fe³⁺ (0.6425 Å) compared to Sb⁵⁺/Fe³⁺ (0.6225 Å) or Sb⁵⁺/Cr³⁺ (0.6075 Å), ^[74] justifying the preferred location into A-B edge- / face-sharing octahedra respectively. The increasing number of unpaired electrons translates into progressively higher magnetic transition temperatures as summarised in Table 3. Magnetic structures are not studied in the LN samples, but the strong magnetic frustration of these systems with stacked honeycomb layers is observed in Mn₂BSbO₆ IL oxides. Stronger d⁶-d⁶ AFM interactions between Mn²⁺ and Fe³⁺ stabilises a high temperature collinear magnetic structure with $k = [0 \ 0 \ 0]$, but an incommensurate helical spin structure is developed at lower temperatures with $k = [0 \ 0 \ k_z]$, where the spins lie in the *ab* plane (Figure 4a-left). ^[59] Weaker d³-d⁵ interactions and higher magnetic frustration index justify the continuous rotation of the spins in a T-dependent k_2 for Mn₂CrSbO₆. ^[12] Coupled magnetoelectric properties were observed in Mn₂FeSbO₆, supported by local lattice effects and suggesting accessibility to new multiferroic materials. [59]

Preliminary studies on impurified Mn_2BSbO_6 with B = Ga and Al ^[75] focus on structural features and show a pressure induced phase transition from IL to DPv type structure. Polymorphism in double corundum (DC) – double perovskite (DPv) *A*-site manganites has only been observed for $Mn_2BB'O_6$ with $B' = d^0/d^{10}$ cations, mainly antimonates, where high temperature has also been proven to trigger the reverse DPv to IL phase transition at room pressure conditions. ^[12]

4) Triple Mn₄B₂O₉ (1:2 B order)

Room pressure polymorphs of $Mn_4B_2O_9$ ($B = Nb^{5+}$, Ta^{5+}) are well known multiferroic materials with space group *P*-3c1. Their composition can also be expressed as $Mn_3MnB'_2O_9$, suggesting a 1:2 order ratio between Mn and *B*' into the *B* sites. Two consecutive (001) honeycomb layers are formed by well-ordered Mn^{2+}/B^{5+} in octahedral coordination, where BO_6 octahedra are always connected via face sharing. One in three *ab* layers is purely formed of MnO_6 octahedra, supporting the 1:2 order in the *B*/*B*' sites of a triple corundum-related structure with ilmenitetype stacking, see Figure 3. The symmetry relation to the corundum structure requires a Λ_2 mode, reflecting the similar relation found between the cubic Pv and the TPv structure of $Mn_3MnB'_2O_9$ through mode Λ_1 . This TC polymorph is recoverable from the high pressure TPv phase by heating above 900 K.^[38]

5) Doubly ordered MnA'BB'O₆ (1:1 A and/or B orders)

Combining cation order into (00I) or (012) layers for *A* and/or *B* sublattices gives rise to *R*3 OIL and NTOtype structures for *A*-site manganites. They show four independent crystallographic sites for cations, thus giving rise to a total of 24 possible permutations when occupied by 4 different cations, but only 6 of those combinations are non-equivalent when considering symmetry arguments. Note that in $Mn_2BB'O_6$ oxides, 2/4 sites are occupied by Mn, thus leaving only three possible arrangements: OIL, OLN (with one existing compound, Co_2InSbO_6)^[57] and NTO type structures, which could also be described as doubly-ordered ilmenite (DOIL) or LiNbO₃-type (DOLN) structures.

Among $Mn_2BB'O_6$ oxides, three compounds are reported to crystallise with OIL structure, *i.e.* Mn_2lnSbO_6 , Mn_2LiReO_6 and Mn_2FeMoO_6 . ^[61,21,62] These are chemically unique for different reasons: Mn_2lnSbO_6 is the only non-polymorphic OIL synthesis product, Mn_2LiReO_6 is the only $Mn_2B^*B'^7+O_6$ and the only Re-based ordered corundum derivative, and Mn_2FeMoO_6 is notable as the only *A*-site manganite stabilised from an irreversible moderate temperature (450 K) cation rearrangement from a different ordered corundum derivative.

According to the expected magnetic frustration, Mn_2LiReO_6 shows no long-range magnetic order, with a spin cluster behaviour dominating the bulk magnetic properties. ^[21] Mn_2InSbO_6 shows three subsequent transitions where a simple ferrimagnetic structure ($k_1 = [0 \ 0 \ 0]$) turns into a *T*-dependent incommensurate helix ($k = [0 \ 0 \ k_2]$) which locks-in at low temperature ($k_3 = [0 \ 0 \ 1/8]$), ^[61] with ~ 120° spin angles between the three independent magnetic sites (see Figure 4b-right), reflecting the 3-fold symmetry of the hexagonal packing as often observed in NTO-type phases. ^[76] Regarding Mn_2FeMoO_6 , a collinear ferrimagnetic structure with all spins lying along the *c* axis with $k = [0 \ 0 \ 0]$ (Figure 4a-middle) is reported, its ordering temperature modulates with the degree of cation order. ^[62]

The more flexible NTO-type structure is reported for nine high pressure oxides with MnA'BB'O₆ general composition. Six of them are $Mn_2BB'O_6$ ($B = Sc^{3+}$, Fe^{3+} , Fe^{2+} , Mn^{2+} and $B' = Sb^{5+}$, Nb^{5+} , Ta^{5+} , Mo^{5+} , W^{6+})^[11,63,64,65,77] and the other three correspond to the $Mn_{3\cdot x}Co_xTeO_6$ (x = 1.5, 2, 2.5) solid solution. ^[14] Among $Mn_2BB'O_6$, the only clear doubly ordered corundum with a chemically based NTO structure as described above are $Mn_2ScB'O_6$ (B' = Sb, Nb, Ta). A site-selective antisite disorder between one Mn and Sc sites induces a weak ferromagnetic moment in all three samples ($T_c = 42, 53$ and 50 K respectively) into the collinear AFM structure, confirmed from neutron diffraction for Mn_2ScSbO_6 . ^[11] Metamagnetism is induced in Mn_2ScTaO_6 for $H_c > 0.5$ T, although the origin of this transition is not clear. ^[77] The accurate cation arrangements for the other three compounds are discussed in the next section.

Despite the presence of paramagnetic cations into all 4 sites, Mn_2FeMoO_6 shows a simple magnetic behaviour, with a collinear $k = [0 \ 0 \ 0]$ structure with spins parallel to the *c* axis, like that of the OIL modification. ⁽⁶³⁾ Mn_2FeWO_6 shows a metamagnetic hysteretic behaviour explained in terms of competing (calculated) collinear structures, though a more complex non-collinear magnetic structure is also suggested. ⁽⁶⁴⁾ A different behaviour is observed from experimental data on Mn_3WO_6 , with an incommensurate conical magnetic structure (partially shown in Figure 4a-right) which satellites are consistent with coexisting $k_1 = [0 \ 0 \ 3/2]$ and $k_2 = [0 \ 0 \ 0.3]$, thus combining the incommensurate and locked features usually observed in honeycomb/triangular lattices. ⁽⁶⁵⁾

Concerning the Mn_{3-x}Co_xTeO₆ (x = 1.5, 2, 2.5) solid solution, ^[14] their magnetic behaviour shows coexisting $k = [0 \ 0 \ k_z]$ and $[0 \ 0 \ 0]$, modulating the magnetic moments into an incommensurate elliptical helix (Figure 4bbottom) in all cases. The coherent evolution of their transition temperatures and the absence of thermal dependence for k_z , in contrast to Co₃TeO₆, is explained in terms of progressively weakened d^6-d^7 superexchange interactions and less effective direct orbital overlap according to GKA rules. ^[39]

Note all these OIL and NTO-type phases, along with the possible OLN arrangement, are polar structures with cation shifts along the *c* axis inducing potential multiferroicity when the magnetic moments lie in the perpendicular *ab* plane (as in helical k_z structures). The spontaneous polarisation and magnetoelectric coupling are estimated for most of them from point charge or DFT calculations, and confirmed by second harmonic generation or pyroelectric current measurements.

В	B'	Struct./S.G.	d _B		d _{B'}	T _N (K)	k ₁	f	Other properties	Ref
			t _{2g}	eg						
Fe	Sb	IL / <i>R</i> -3	3	2	10	260	[000]	1.8	Giant MEC at T_{N1} k ₂ = [000.07] T_{N2} = 50 K Relaxor AFE < 650 K	59
Cr	Sb	IL / <i>R</i> -3	3	0	10	60	[00k _z]	3.3	T-dependent kz	12
Sc	Sb	NTO / <i>R</i> 3	0	0	10	42	[000]	3.3	wFM from site selective antisite	11
In	Sb	OIL / <i>R</i> 3	5	5	10	38	[000]	4.2	<t<sub>N2 = 17 K: [00kz] T_{N3} = 8 K: lock-in [0 0 1/8]</t<sub>	61
Sc	Nb	NTO / <i>R</i> 3	0	0	0	53		3.9	Insulator	77
Sc	Та	NTO / <i>R</i> 3	0	0	0	50		3.9	Insulator. Metamagn. H _C = 0.5 T	77
Mn	W	NTO / <i>R</i> 3	3	2	0	58	[0 0 3/2] + [0 0 0.3]	4.8	$T_{N2} = 20$ K: $[00k_z]$ conical	65
Fe	W	NTO / <i>R</i> 3	3	2	1	75	[000]		Colinear // c	64
Fe	Nb	LN / <i>R</i> 3c	3	2	0	90			Polar	60
Fe	Та	LN / <i>R</i> 3c	3	2	0	80		2.6	Polar	60
Fe	Мо	NTO / <i>R</i> 3	3	2	1	337	[000]	0.7	Colinear // c	63
Fe	Мо	OIL / <i>R</i> 3	3	2	1	229	[000]		Colinear // <i>c</i>	62
Li	Re	OIL / <i>R</i> 3	0	0	0	46		4.3	Cluster spin glass	21
A/A'	B/B'	Struct./S.G.	d _A	d_B	d _{B'}	T _N (K)	k 1	f	Other properties	Ref
Mn _{1.5} Co _{0.5}	CoTe	NTO / <i>R</i> 3	5/7	7	10	67	[0 0 0.080]		Elliptical helix	14
MnCo	CoTe	NTO / <i>R</i> 3	5/7	7	10	72	[0 0 0.094]		Elliptical helix	14
$Mn_{0.5}Co_{1.5}$	CoTe	NTO / <i>R</i> 3	5/7	7	10	66	[0 0 0.117]		Elliptical helix	14

Table 3. Main magnetic features of $Mn_2BB'O_6$ DC phases. The lower part of the table shows compounds in the solid solution $Mn_{3,x}Co_xTeO_6$ with mixed A sites.

* Missing values imply the compound is not fully studied



Figure 4. Representative examples of all types of reported magnetic structures for corundum related high pressure *A*-site manganites. a) [110] projection of the nuclear unit cell showing representative examples with the spins lying on the *ab* plane (left) or along the *c* axis (middle). Right panel shows part of the conical magnetic structure only reported for Mn_2MnWO_6 . b) top view of magnetic structures with commensurate propagation vector (top) and consecutive spin angles 0°, 0° < θ < 180° and 120°. Bottom panel shows incommensurate structures with circular and elliptical helix models. Figures adapted from references [11, 14, 61, 62,65].

C) Structural aspects to reconsider.

From the chemical point of view, the cation distribution in Mn_2FeMoO_6 and Mn_2FeWO_6 , with alternating (012) Mn and 1:1 Fe:Mo(W) layers, ^[63,64] actually describes an OLN structure. It is reasonable that OLN and OIL modifications are very close in energy, enabling their formation in similar synthesis conditions or their

interconversion under soft conditions, as observed in Mn_2FeMOO_6 , ⁽⁶²⁾ in simple $MnBO_3$ IL to LN polymorphs or the OIL to OLN in Co_2InSbO_6 , triggered by thermal relaxation. ^[57] In the case of Mn_2FeWO_6 , the authors highlight that Mn/Fe antisite is not well characterised in the preliminary study, and this could involve "an effect on the magnetic properties". In the reported structure, antisite exchange of Fe with the so-called Mn1 or Mn2 sites would turn the OLN into an actual NTO or an OIL phase, respectively. Only a full structural characterisation from neutron diffraction will answer this ambiguity and reveal the magnetic structures that may explain the bulk magnetic behaviour of this phase. Note Mn_2ScNbO_6 and Mn_2ScTaO_6 , also reported as NTO-type structures, show up to 40% antisite disorder between Sc and one Mn, thus lying in the limit between NTO and OIL structures. ^[77] The reported field induced transition is likely due to different magnetic domains with similar energy in inhomogeneous samples.

 Mn_3WO_6 is reported as NTO-type, this is a comprehensible mistake, considering that Ni_3TeO_6 itself shows a similar OIL cationic arrangement.^[58] Regarding the $Mn_{3-x}Co_xTeO_6$ solid solution, different proportions of mixed occupancies involve the intrinsic double order into the NTO-type structure, even though a clear site preference is observed for Mn in the Te (00I) layer, suggesting the OIL is more stable than the OLN arrangement in these compounds.

D) Perspectives.

HPHT synthesis is confirmed from all previous materials as an essential tool to access many new polymorphic compounds among A-site manganites, with different types of cation order and physical properties. Some of them deserve a special attention regarding future possibilities:

1) Mn₂O₃ is remarkable as the most polymorphic among *A*-site manganites: starting from the α-Mn₂O₃ (S.G. *Ia*-3) room pressure bixbyite phase, it transforms at low temperature (T < 300 K) to β-Mn₂O₃ (S.G. *Pbca*) and the metastable γ-Mn₂O₃ (S.G. *I*4₁/*amd*) obtained by soft methods. It presents three high pressure transformations to the ε-Mn₂O₃ (Ilmenite type, S.G. *R*-3) between 12 < P < 17 GPa and 1200 < T < 1700 K. It also transforms to ζ-Mn₂O₃ (QPv type, S.G. *Im*-3) between 18 < P < 23 GPa and 900 < T < 1500 K and to a post-perovskite structure above 28 GPa, δ-Mn₂O₃ (S.G. *Cmcm*). Under extreme pressure conditions, unquenchable θ- Mn₂O₃ (orthorhombic) and η- Mn₂O₃ (perovskite S.G. *Pnma*) coexist with the post-perovskite structure. Such a rich P-T phase diagram in the simplest chemical composition among A-site manganites suggests many new HPHT phases could be found for more complex chemical compositions.

2) $Mn_{11}Ta_4O_{21}$ [^{78]} was found as an overstoichiometric ($Mn_8(Mn_3Ta_4)O_{21}$) hexagonal perovskite (HPv) with a 14H-(*cchchch*)₂ sequence, similar to those of the related 14H high pressure HPvs BaVO₃ and BaCrO₃.^[79,80] At T_{N1} = 88 K AFM ordering develops and below T_{N2} = 56 K a spin reorientation originates a ferromagnetic component on additional *A*-site Mn cations, also inducing a magneto-structural coupling with possible magnetoelectric effect. The finding of this first hexagonal A-site manganite (HAM) suggests the accessibility to other related compounds. New HAM samples may be found and HPHT phase transitions to cubic polymorphs can be expected, as observed in other HPvs. [⁸¹] Moreover, the close structural relation between $Mn_4B_2O_9$ and $Mn_3MnB'_2O_9$ structures, as a TC (with hexagonal close packing) and TPv structures with 1:2 *B*-site order suggests many new compounds with TC, TPv and HPv structures may be accessible through HPHT synthesis.

3) Mn_3TeO_6 ^[13] and $Mn_3MnB'_2O_9$ ^[37,38] oxides have shown interconversion between their respective room and high-pressure polymorphs. In both cases, a reduced band gap was observed for the HP phase, due to the lower connectivity of $B'O_6$ octahedra. Band gap tuning is thus expected for other polymorphic compounds, allowing for manipulation of their optical properties.

4) CaMnMnWO₆ DPv ^[24] shows a spin glass behaviour from *B*-site geometric frustration. This finding suggests accessibility to several new compounds with frustrated magnetic properties never observed among *A*-site manganites. For instance, substitution of Mn²⁺ (S = 5/2) by Cu²⁺ (S = $\frac{1}{2}$) may induce quantum spin liquid behaviour.

5) Among cation ordered corundum related phases, OLN type structure has not yet been observed for *A*site manganites. However, it has been recently reported for the related Co-based HPHT Co₂InSbO₆. ^[57] Under HP conditions, this compound adopts a new *R*3 structure deriving from cation order of a never observed hypothetical *R*32 corundum derivative, where *A*/*A*' cations locate at face-sharing octahedra. A two-step phase transition is triggered by temperature under room pressure conditions, giving rise to two new OLN phases. This finding, together with precedents of high temperature phase transitions in polymorphic *A*-site manganites and a rich history of similar transitions for other HPHT materials, suggests accessibility to OLN polymorphs for Mn₂*BB*O₆ samples. Other cationic rearrangements may also be observed from high temperature phase transitions of known OIL phases into new OLN or NTO structures. This will of course produce dramatic changes in their magnetic properties.

6) Interplay between corundum- and perovskite – related polymorphs is common among *A*-site manganites with d^0 or d^{10} *B'* cations as a function of pressure and temperature conditions, for instance Mn_{3-x}Co_xTeO₆, ^[13,14] Mn₂BSbO₆, ^[11,12,15,59] and Mn₂LiReO₆. ^[21] Temperature control has also proven essential to induce cation orderdisorder phase transitions between DPv and DDPv structures, as in *R*MnMnTaO₆. ^[24] Cation rearrangement also gives access to phase transitions between different corundum-related superstructures, triggered by thermal relaxation in Mn₂FeMoO₆. ^[62] Small compositional modifications in Mn₂FeMoO₆ allowed the stabilisation of a perovskite-related polymorph for Mn₂(Fe_{0.6}Mo_{0.2})MoO₆. ^[22] This single case, in the middle of a rich playground for polymorphism, suggests accessibility to several new functional oxides with tuned properties from slight chemical modification.

7) MnTaO₂N is the single example of a HP A-site manganite oxynitride. ^[62] Other polyanionic A-site manganites should be expected, allowing for tunability of cation charge and thus inducing further degrees of freedom for potential spin, charge and orbital orders in new multifunctional materials with modulated spin (SDW) and charge density wave (CDW) structures.

In conclusion, new unknown polymorphs may be found for *A*-site manganites under HPHT conditions, such as hexagonal perovskites, different combinations of order into *A* and *B* sublattices in new DDPv structures, combined cation order into $AMn_3B_2B'_2O_{12}$ QDPvs or $Mn_2AMnB'_2O_9$ DTPv, layered or oxygen deficient perovskite phases, including brownmillerite-type, Ruddlesden-Popper, Dion-Jacobson or Aurivillius type structures. Among corundum related phases, new cationic rearrangements are also possible, with unknown OLN, *R*32 and its ordered derivatives yet to be found. Polymorphism, only found among $B' d^0/d^{10}$ cations up to date is also expected to provide new compounds among these oxides. Further compositions with mixed anions are also expected. The already success on finding new materials with a variety of properties and compositions advocates to continue the search on this prolific playground of HPHT *A*-site manganites where many new compounds are envisaged.

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