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Adeline Andrieu, Jean-François Brillhac, Gontrand Leysens, Cornelius Schonnenbeck, Samuel Jeanjean, Fabien Halter, Christian Chauveau, Youssef Berro, Marianne Balat-Pichelin, Clément Dumand

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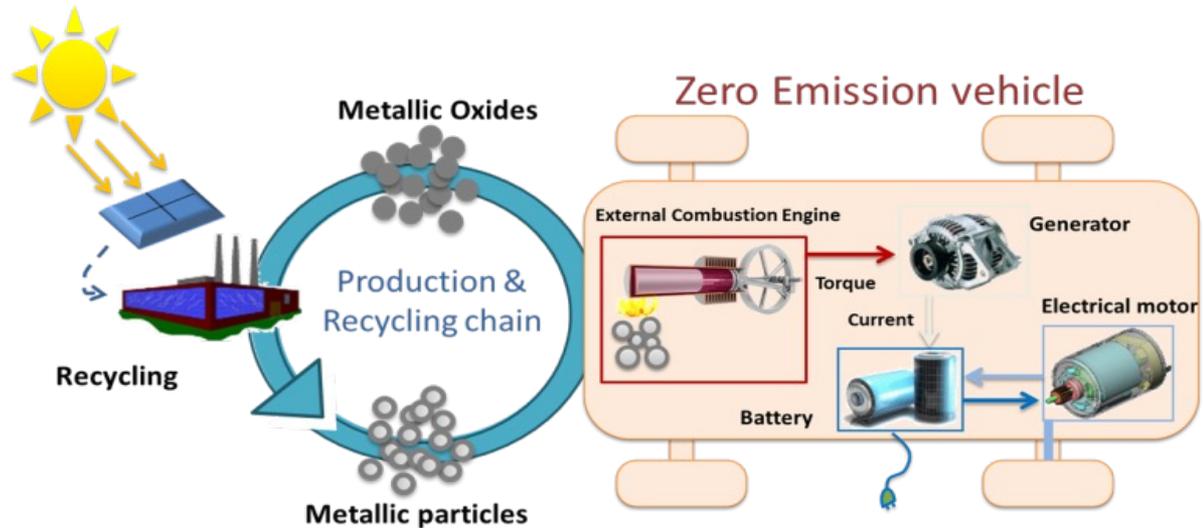
# Combustibles métalliques durables pour la mobilité du futur

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Samuel JEANJEAN<sup>2</sup>, Fabien HALTER<sup>2</sup>, Christian CHAUVEAU<sup>2</sup>,  
Youssef BERRO<sup>3</sup>, Marianne BALAT-PICHELIN<sup>3</sup>,  
Clément DUMAND<sup>4</sup>

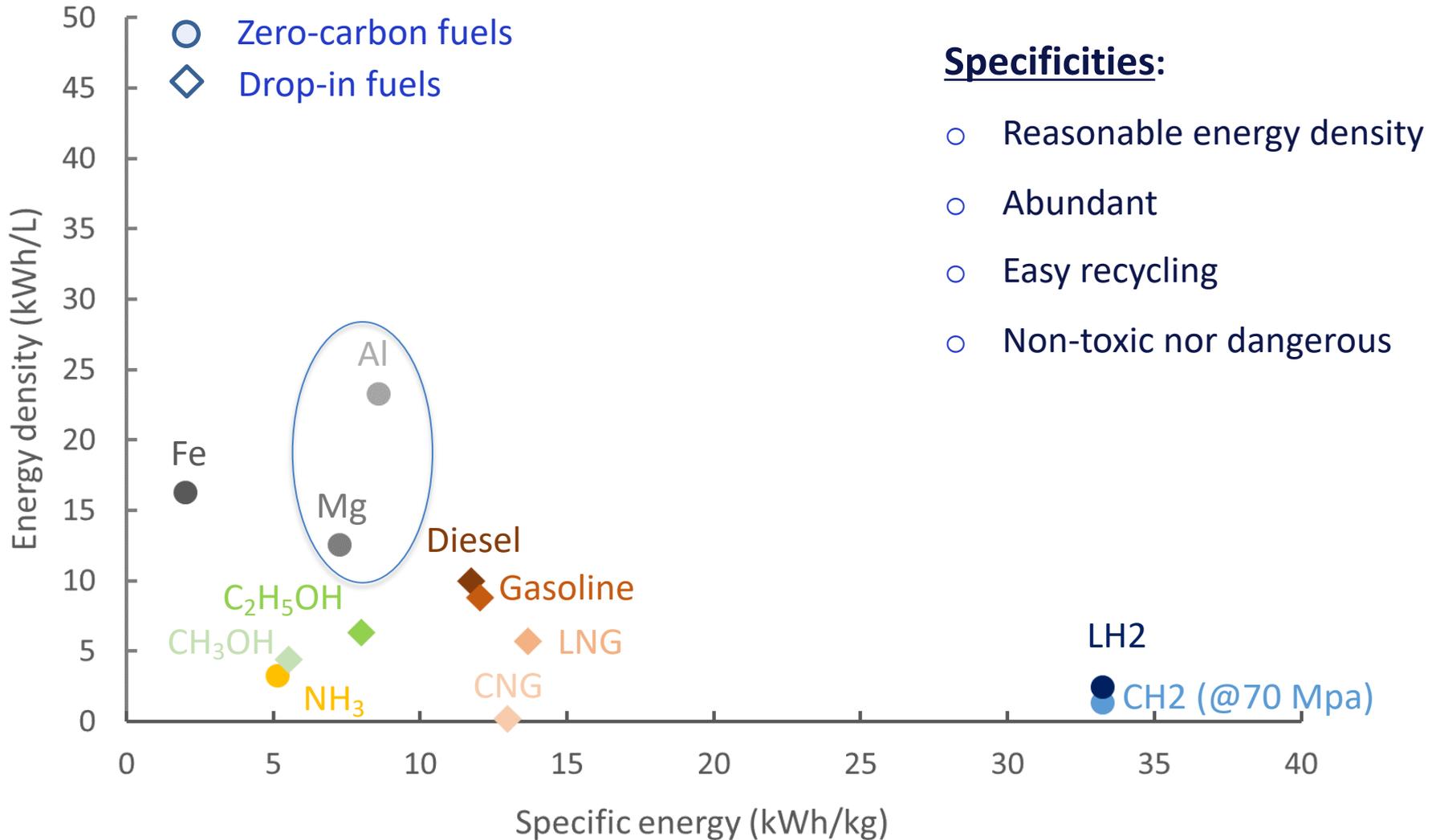
<sup>1</sup>LGRE-UHA, UR 2334, <sup>2</sup>ICARE-CNRS, UPR 3021, <sup>3</sup>PROMES-CNRS, UPR 8521, <sup>4</sup>STELLANTIS

**SFGP 2022, Toulouse, 7-10 novembre 2022**

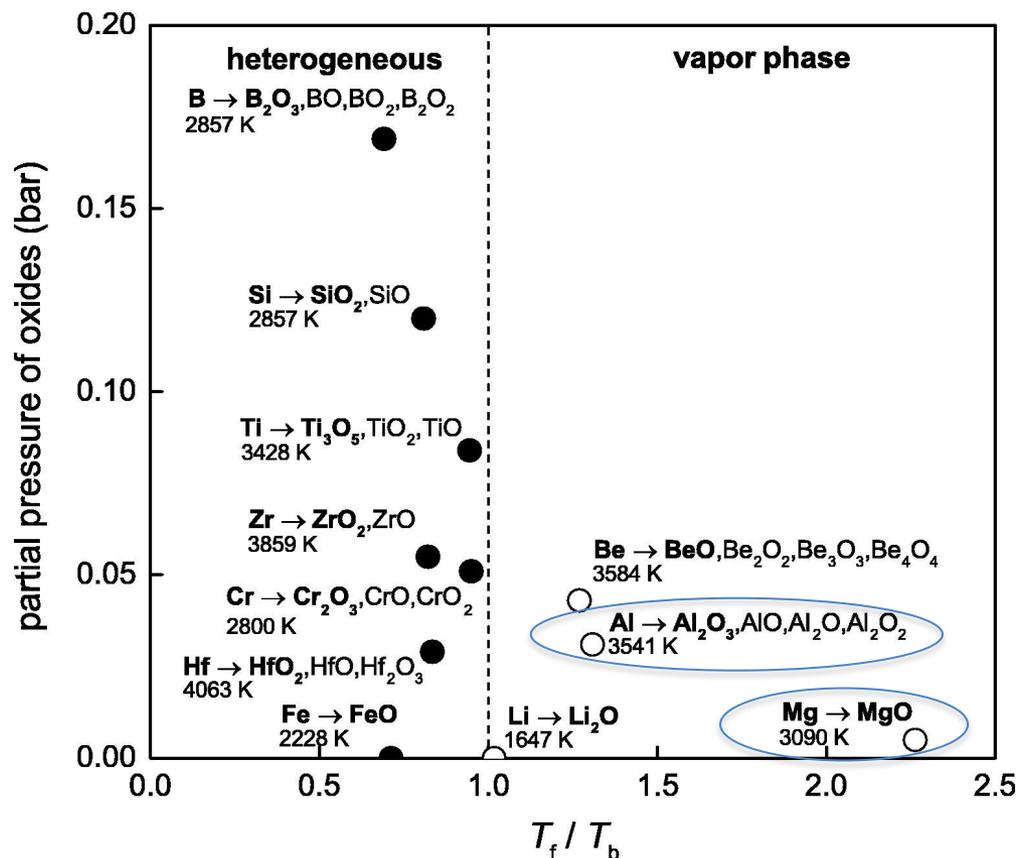
# New energy system for vehicles



# Metal selection

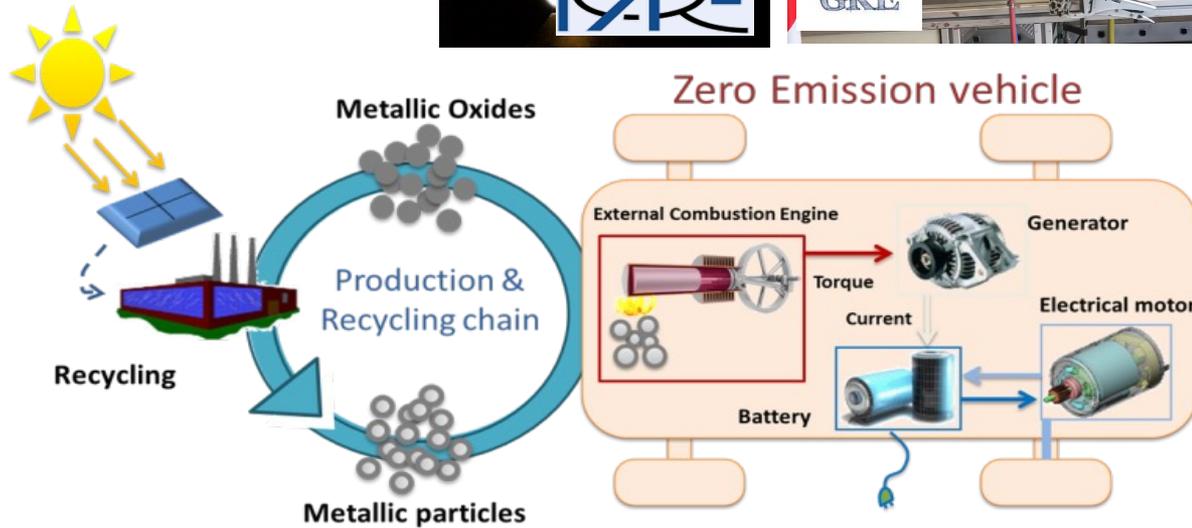
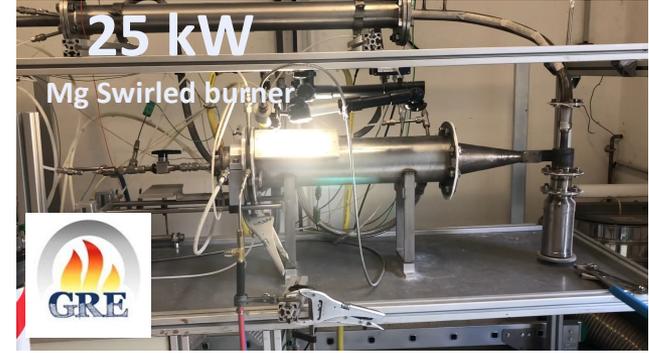


# Metal particle combustion modes



Credits: Bergthorson et al.<sup>1</sup>

1: Bergthorson JM, Goroshin S, Soo MJ, Julien P, Palecka J, Frost DL, et al. Direct combustion of recyclable metal fuels for zero-carbon heat and power. *Applied Energy*. 2015;160:368-82.

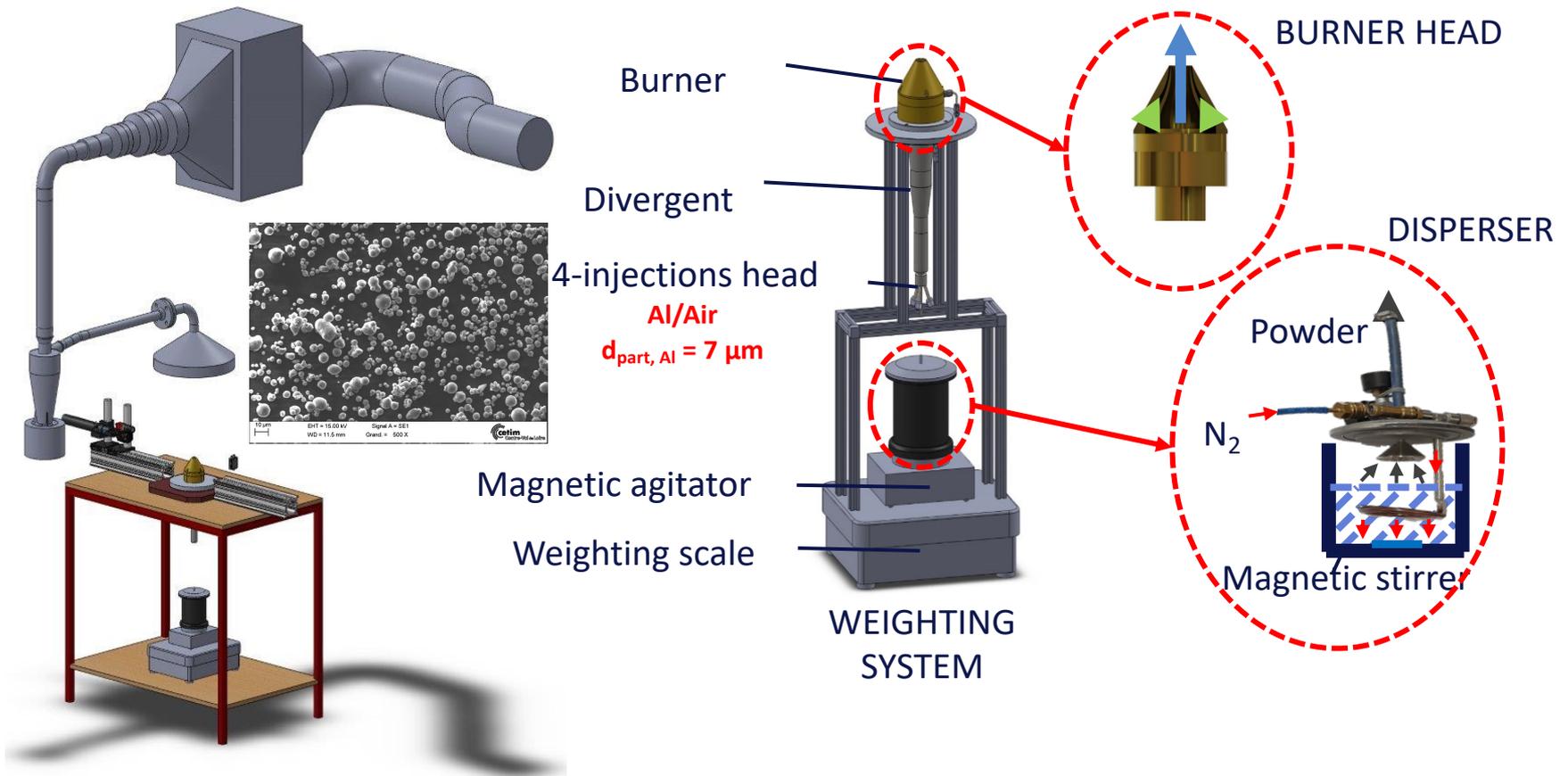


# Objectives of the STELLAR Project

- (i) Flame stability
- (ii) Heat recovery
- (iii) Mitigation of NO<sub>x</sub> emissions
- (iv) Oxide trapping
- (v) Recycling

# Bunsen-type burner (ICARE)

in order to obtain a laminar, conical Al/air flame



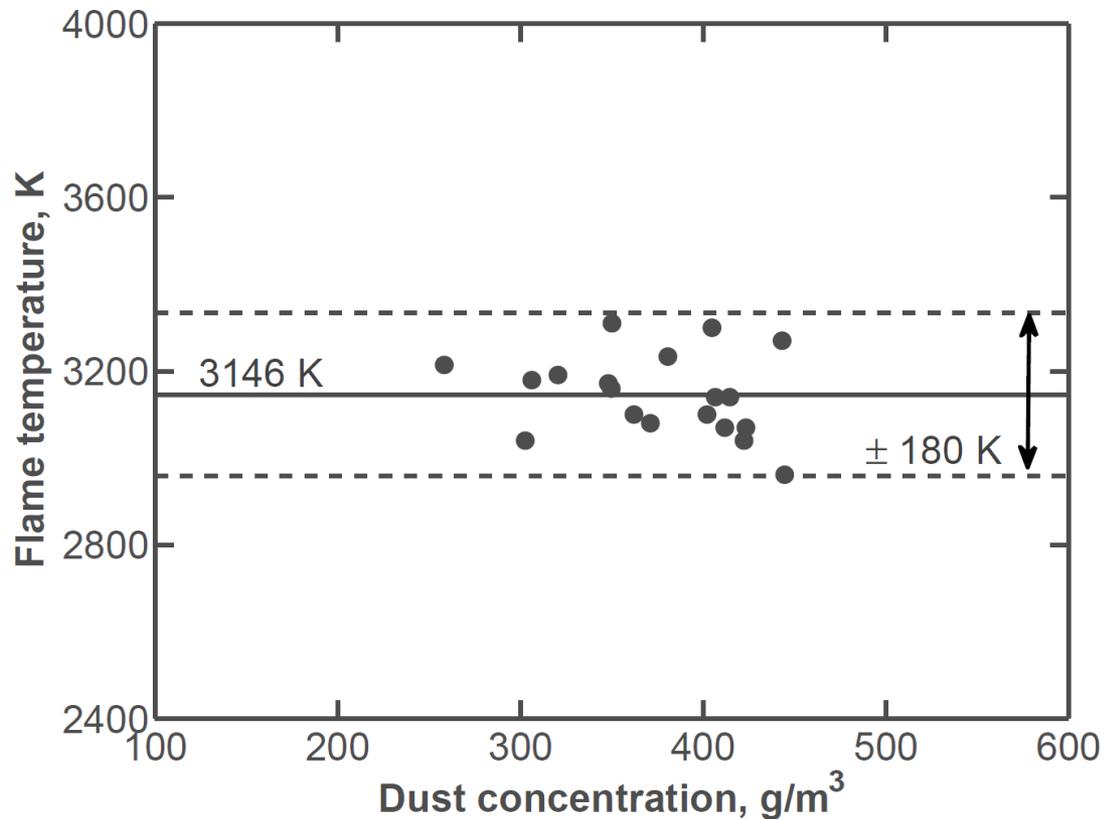
# Laminar Al/air flame



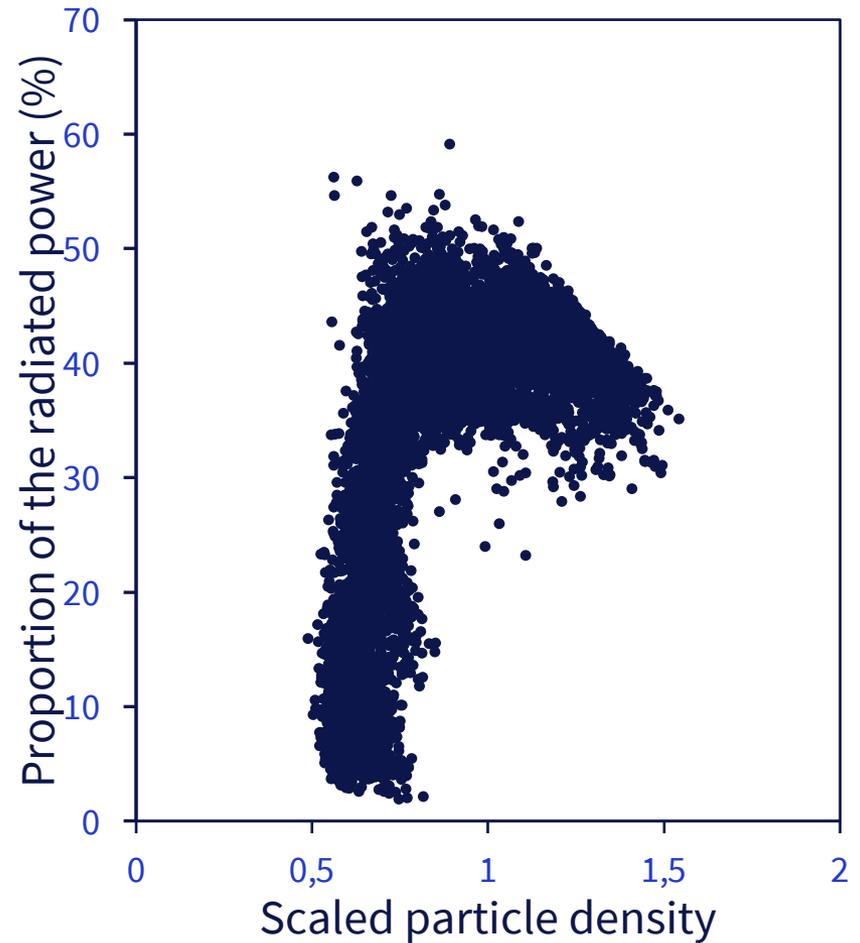
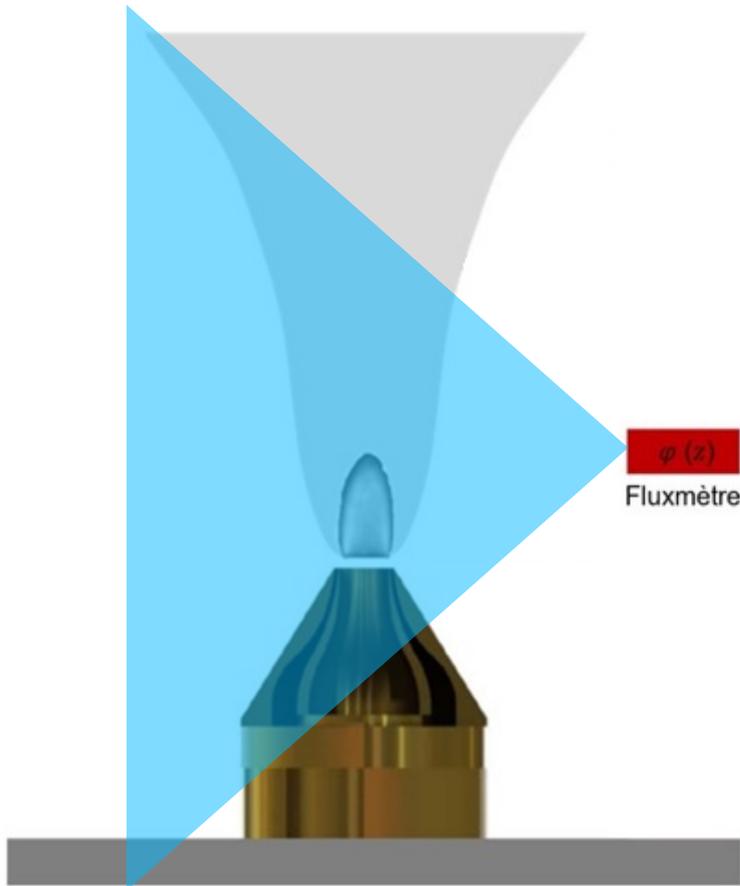
# Gas phase temperature (Al/air flame)

For  $0,8 < \phi < 1,5$  , flame temperature remains constant.

- Adiabatic flame temperature: 3540 K.



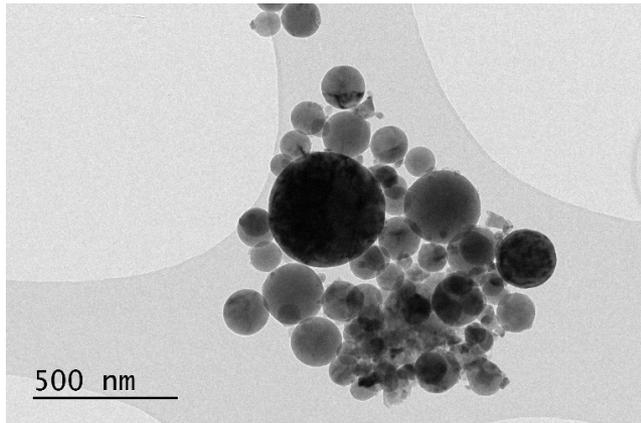
# Flame radiation (Al/air flame)



- Radiation plays an extremely important role.

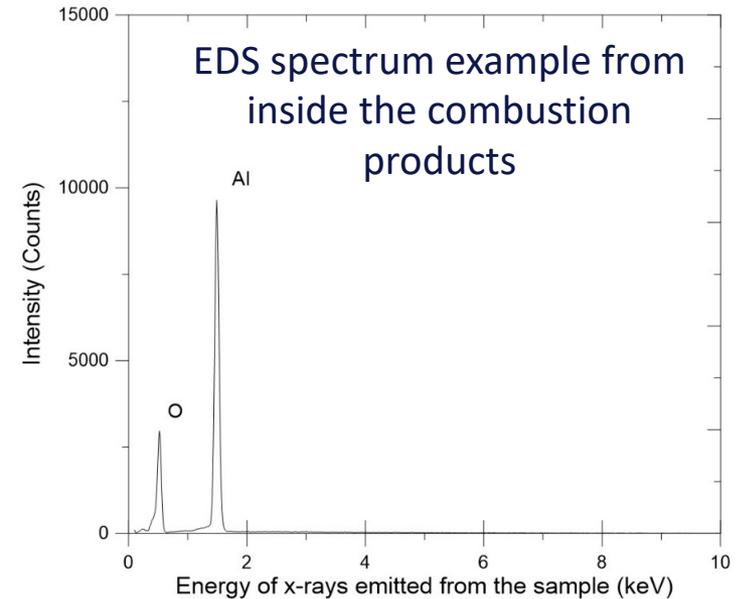
# Condensed products (from Al/air flame)

N<sub>2</sub>



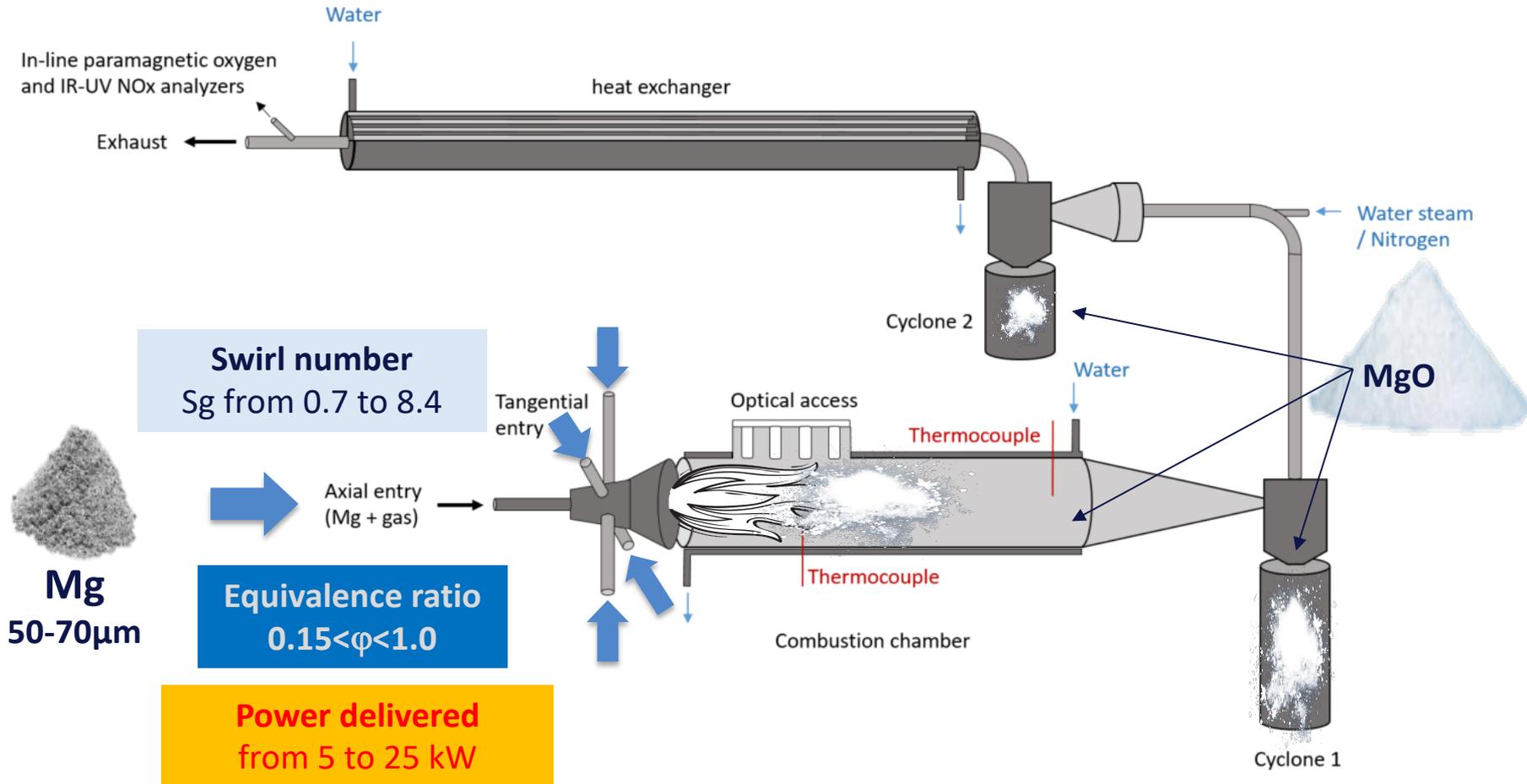
- TEM image of the nanoscale condensed combustion products

- Mean diameter of 110 nm

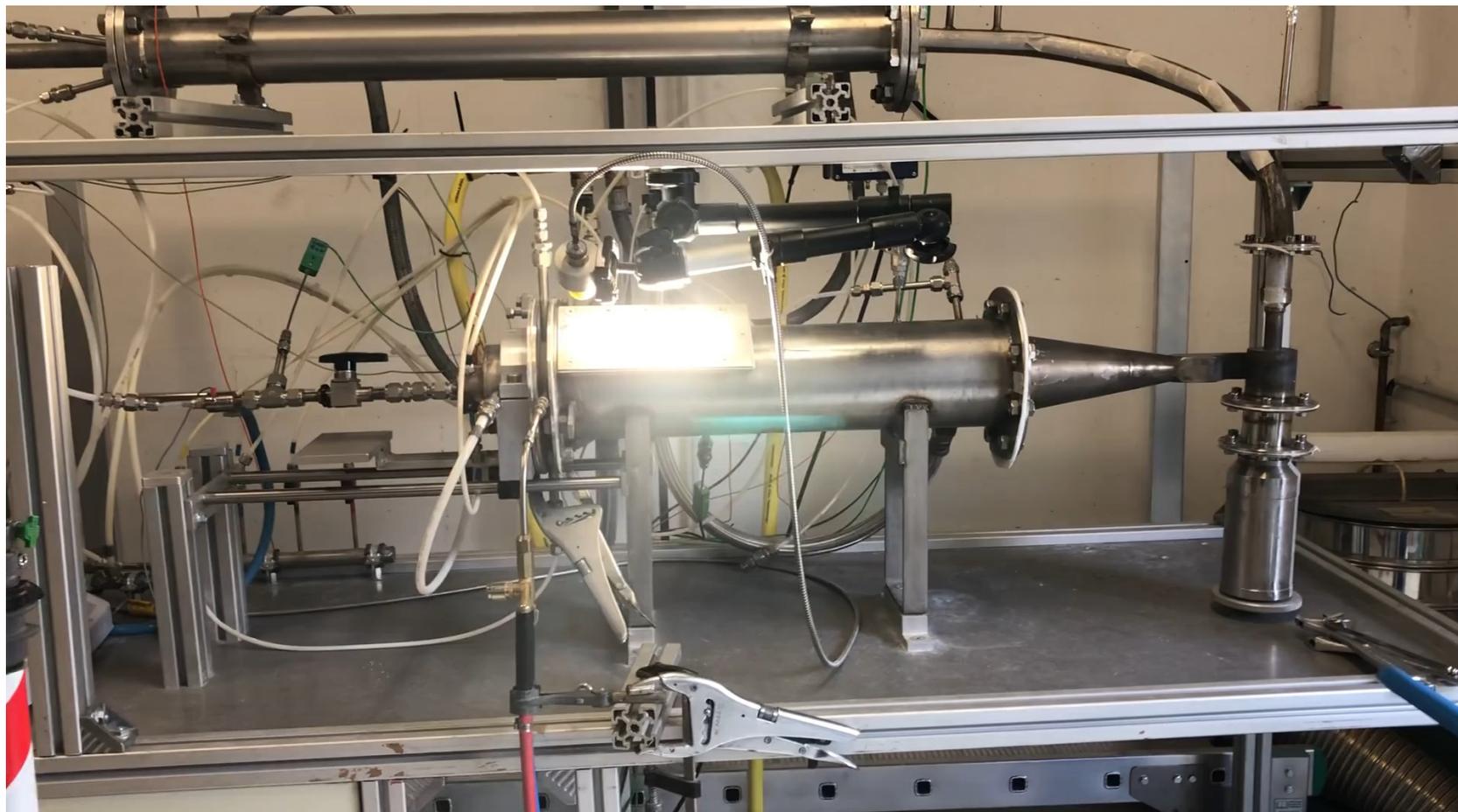


- Combustion efficiency close to unity

# Mg/Air combustion system (LGRE)

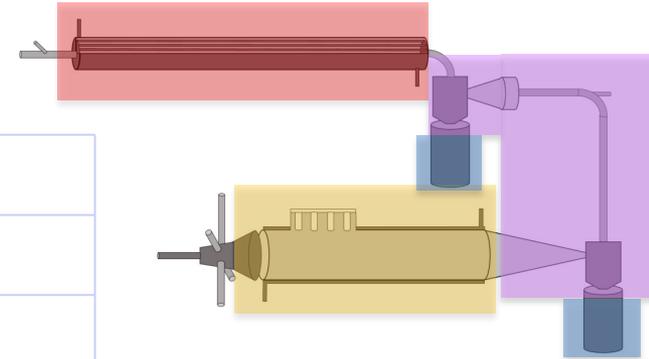
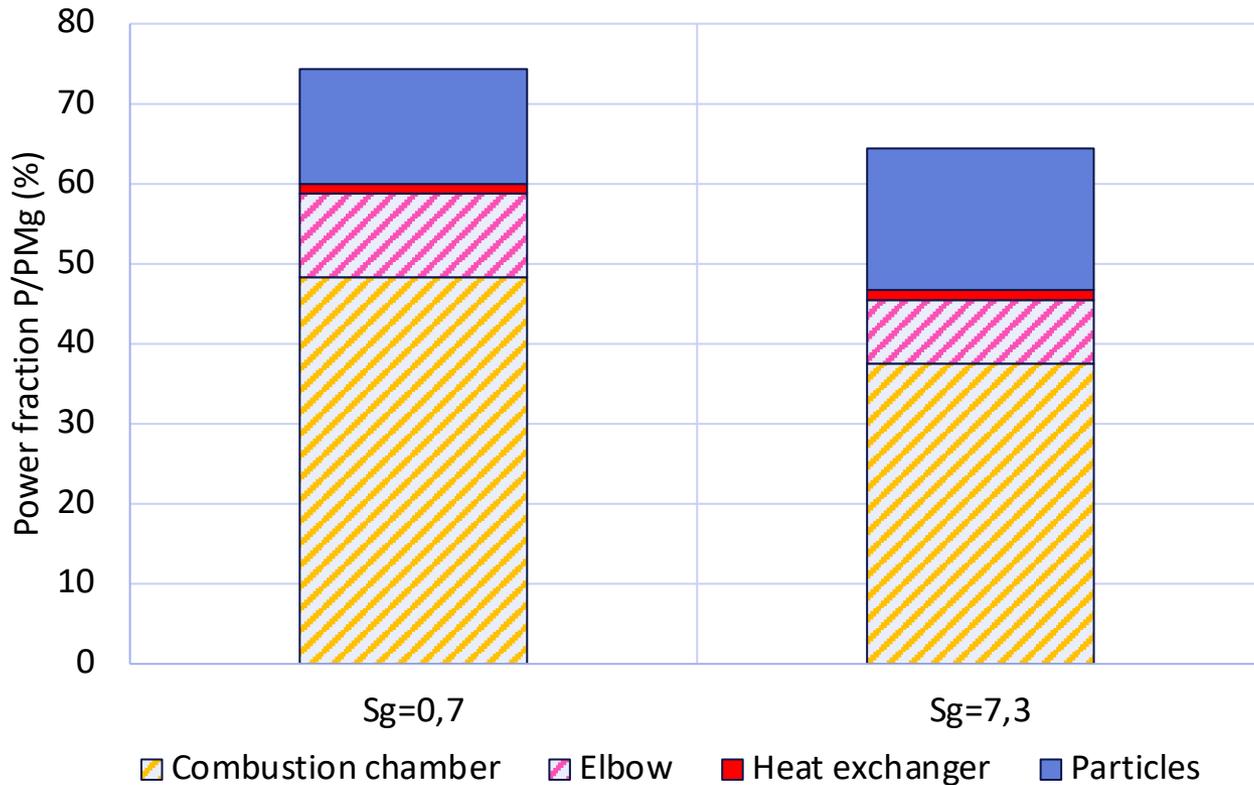


# Mg/Air combustion system (LGRE)

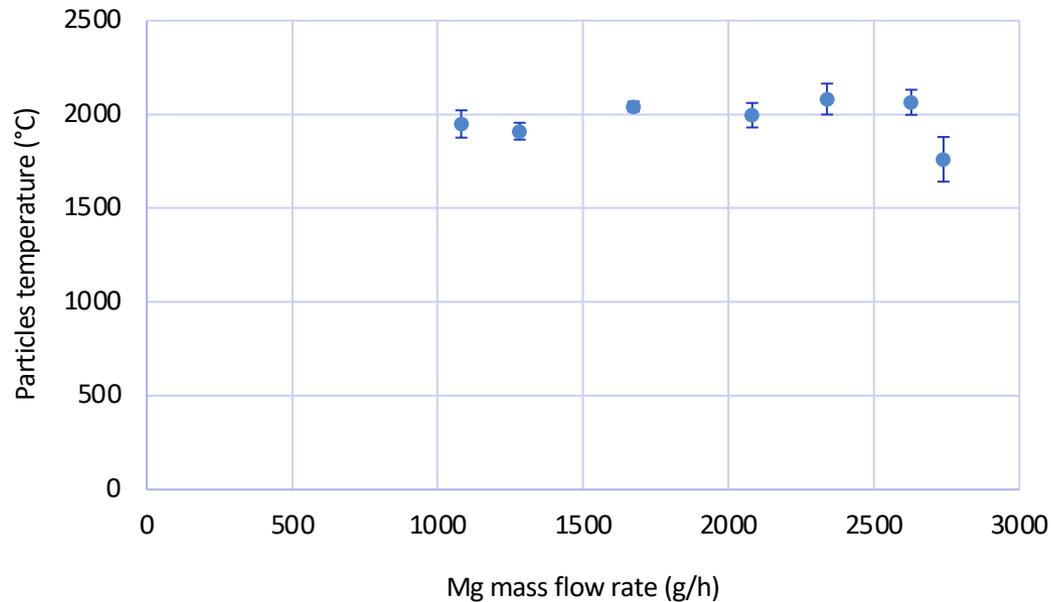
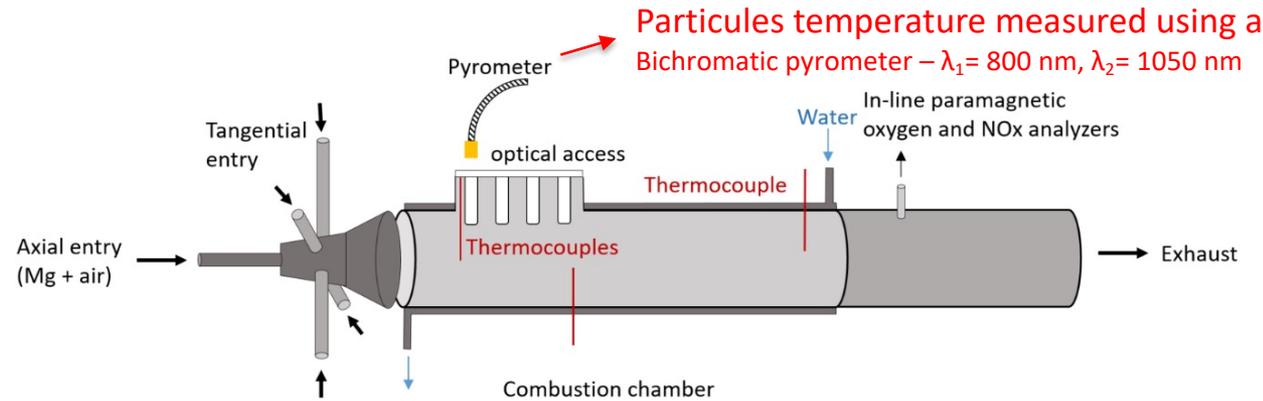


# Mean Power distribution (Mg/air flame)

for  $Sg=0.7$  and  $Sg=7.3$

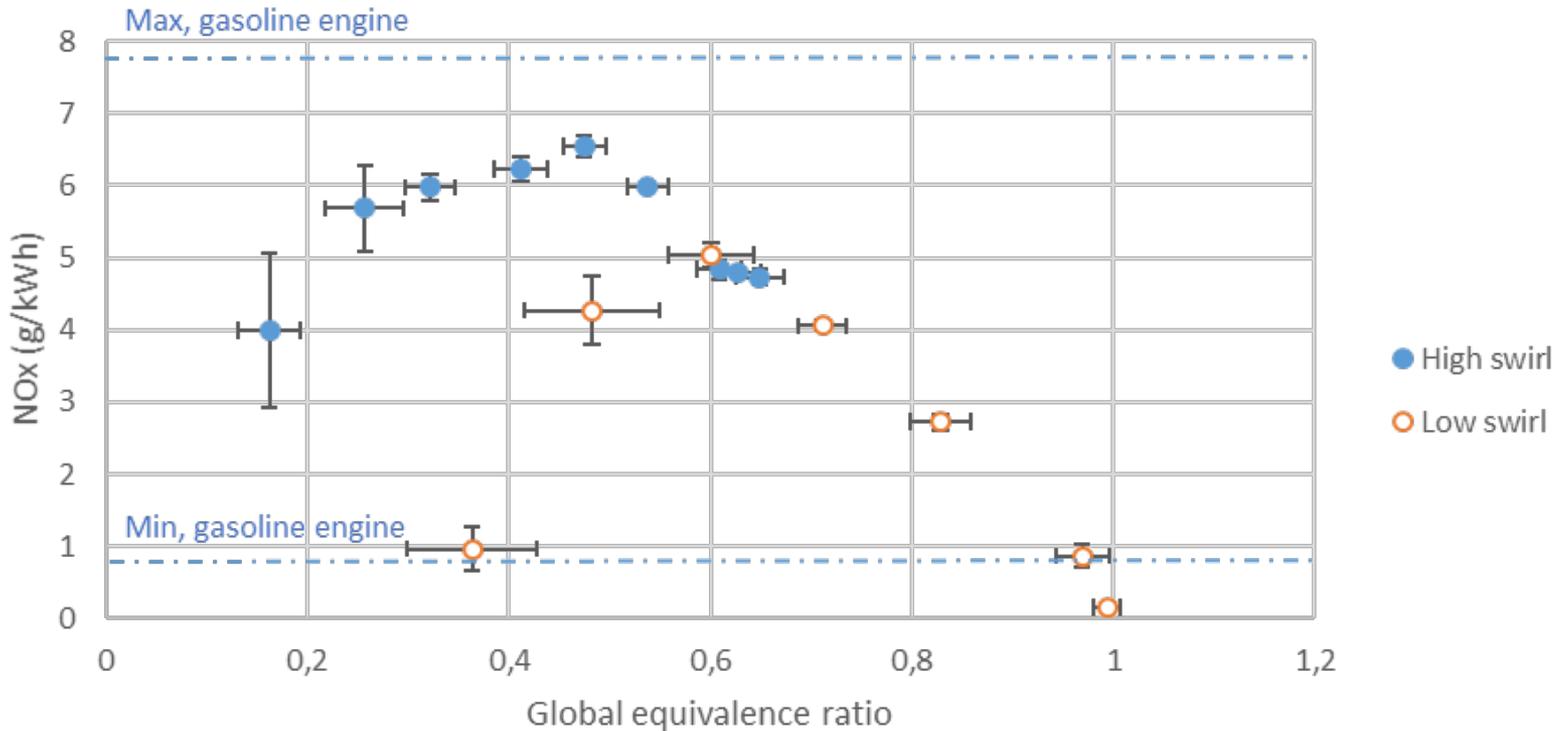


# Particles temperature (Mg/air flame) for $S_g=7.3$



- The observed temperature is below the flame adiabatic temperature (water-cooled walls, cooled MgO screen).
- Particles temperature is not influenced by the global equivalence ratio.

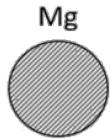
# NOx emissions (Mg/air flame) for $S_g=0.7$ and $S_g=7.3$



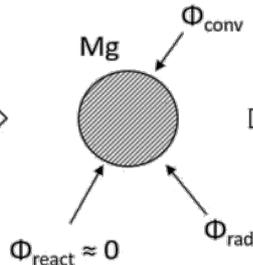
- Level of NOx emissions lower than that from a classical ICE engine

# Mechanism proposed for Mg oxidation and subsequent NOx production

Injection  
of Mg particles at  
ambient temperature

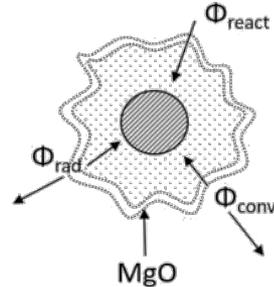


Heating-up  
and ignition  
(by radiation and  
convection)



Slow oxidation  
of Mg by O<sub>2</sub>

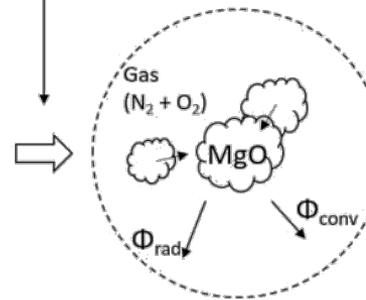
Vaporisation of Mg  
and Oxidation  
of Mg in gaseous phase  
(duration ≈ 6 ms)



O<sub>2</sub> is consumed by Mg  
oxidation  
(2 Mg + O<sub>2</sub> → 2 MgO)  
MgO temperature up to  
2000/3000 °C

Condensation  
of MgO

NOx formation starts when Mg is totally oxidized



$T_{MgO} \approx 2000/3000$  °C  
Gas surrounding the particles  
is heated-up by MgO  
O<sub>2</sub> concentration depends on  
O<sub>2</sub> consumed by Mg oxidation

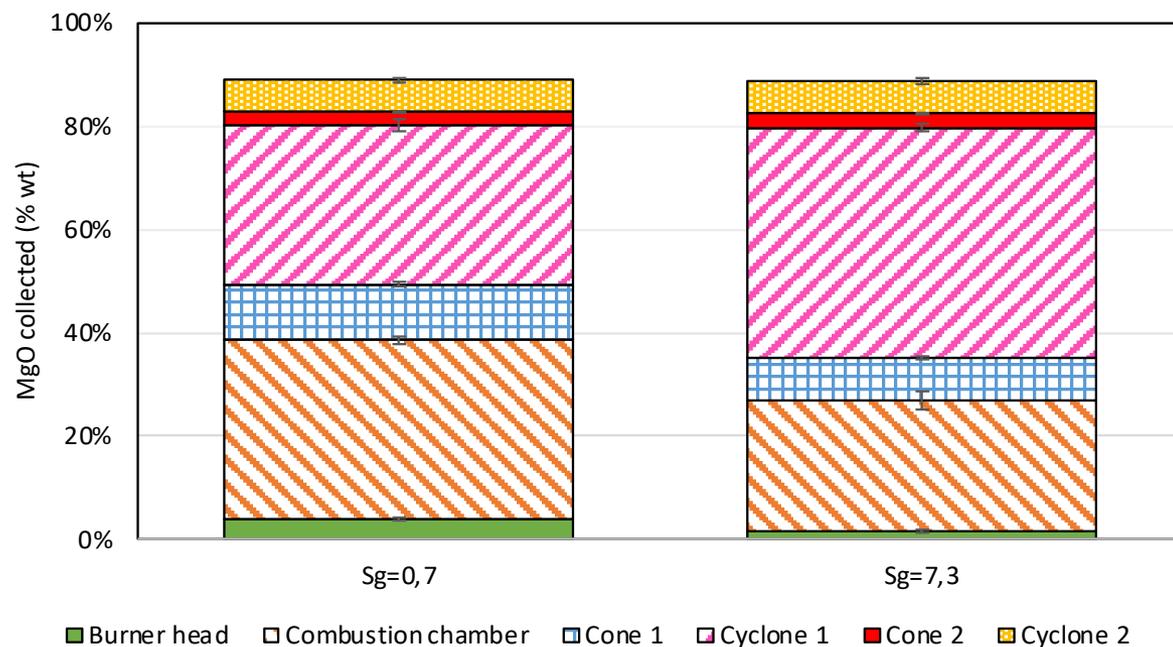
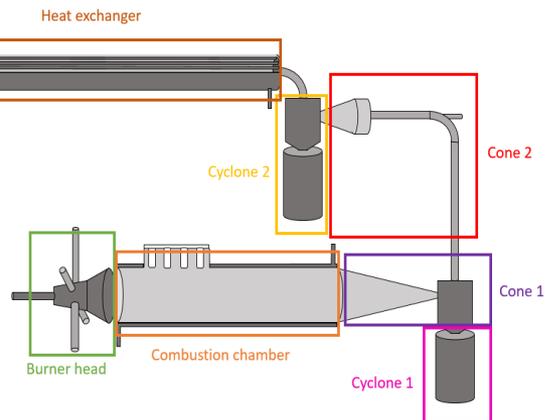
NOx formation



Part of the remaining  
O<sub>2</sub> is consumed for  
NOx yielding

# Mass fraction distribution of collected MgO

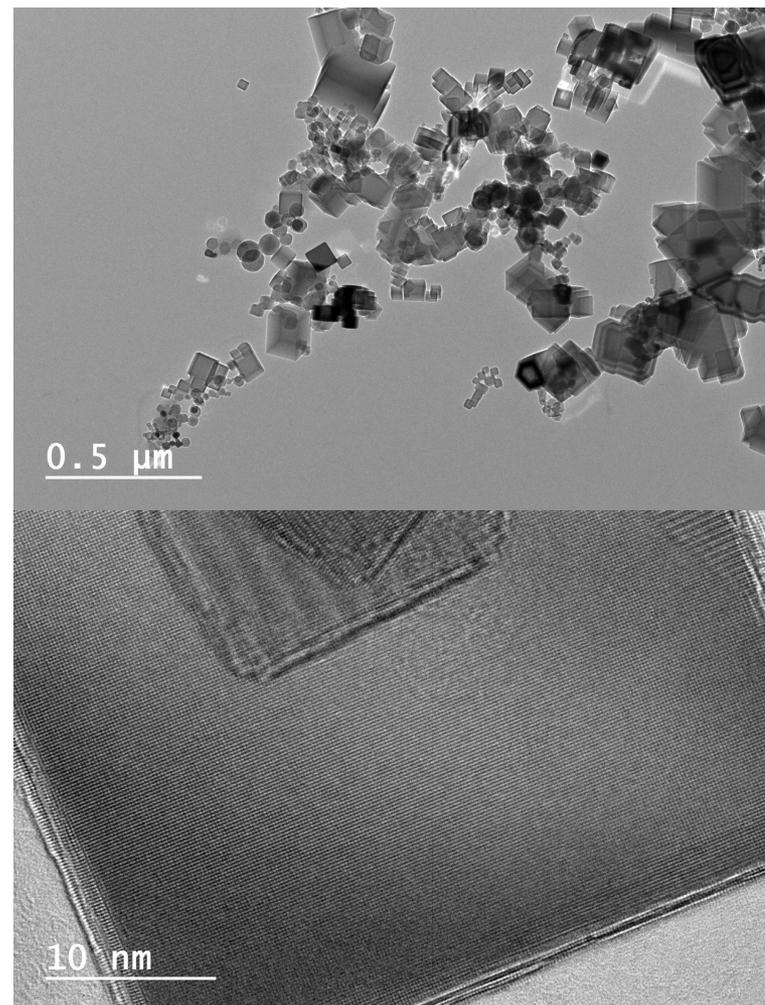
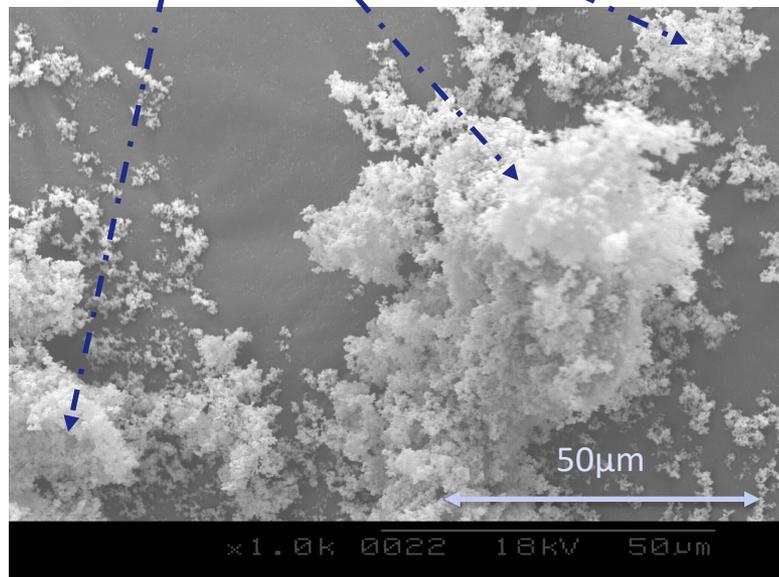
for  $Sg=0.73$  and  $7$



- More than 90 %wt of MgO is trapped in the system **(for regeneration)**
- **Combustion efficiency close to unity**

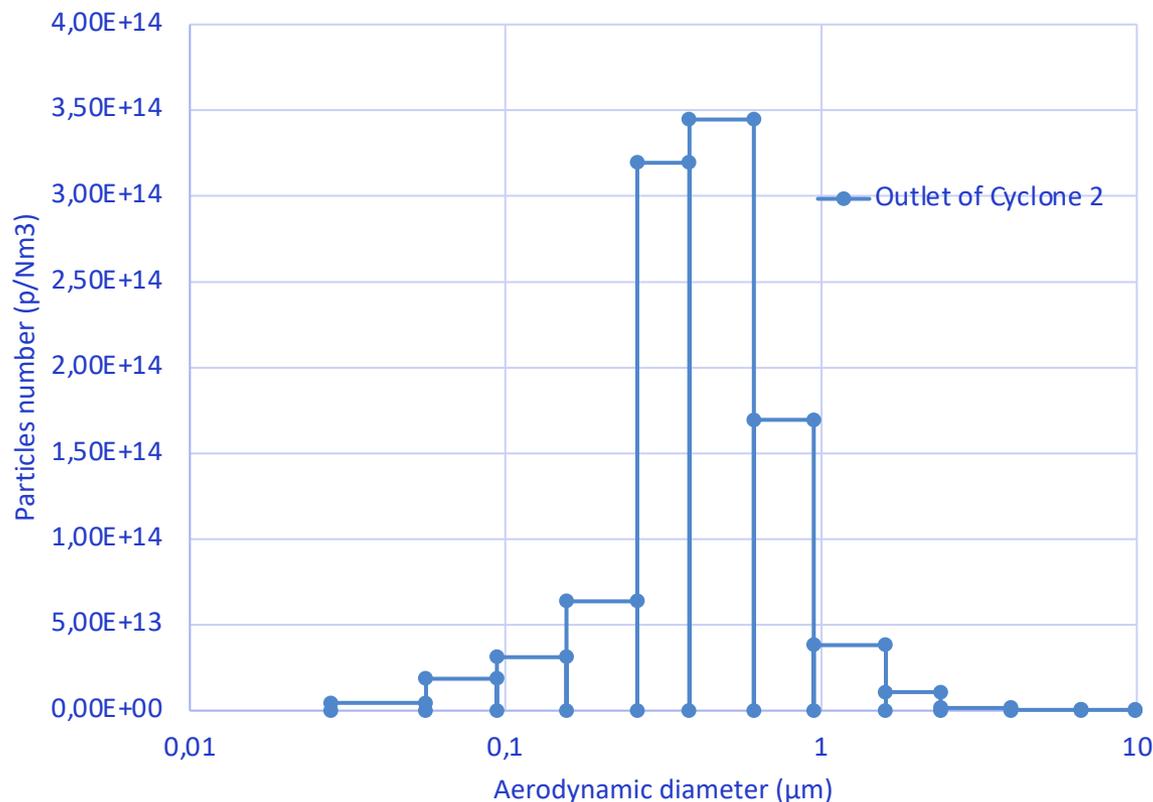
# SEM and TEM images of condensed MgO particles for $S_g=0.7$

Agglomerates of aggregates  
of MgO crystals



# Size distribution of MgO particles in the exhaust gas

(ELPI measurements at the outlet of cyclone 2,  $S_g=0.7$ ,  $\Phi=0.6$ )



- Number size distribution of the MgO particles at the cyclone outflow is widespread in the range 100nm-1 $\mu\text{m}$ .

# Conclusions and perspectives

The available data prove that metal fuels can be burned in power devices at nearly-**equivalent power densities** to modern hydrocarbon-fuelled systems.

NOx emissions can be controlled and more than 90%wt of the MgO produced is trapped in the system.

Metal fuels can be used directly for **heating applications** (high temperatures) or for producing **mechanical** and **electrical power**.

Metal powders could be potentially **retro-fitted** in coal power stations, or those fuelled by biomass.

# Thanks for attention